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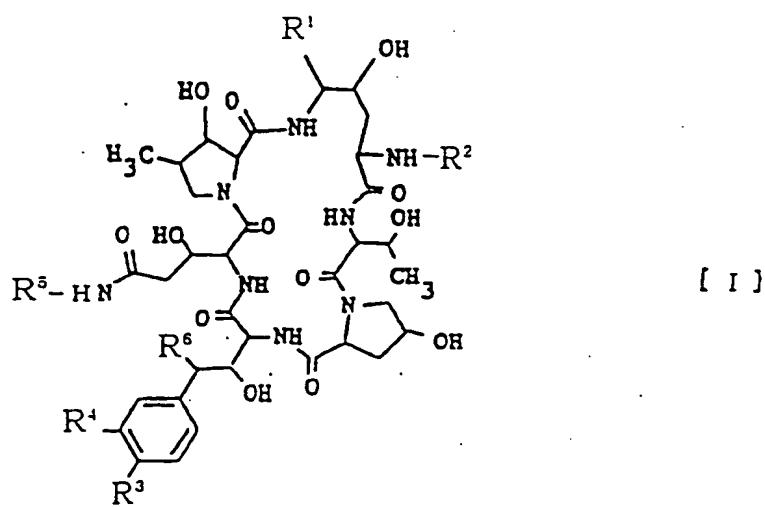
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(52) Cyclic antimicrobial peptides and preparation thereof.

(57) A polypeptide compound of the following general formula :



wherein

R¹ is hydrogen,
R² is acyl group,
R³ is hydroxy or acyloxy,
R⁴ is hydroxy or hydroxysulfonyloxy,
R⁵ is hydrogen or lower alkyl which may have one or more suitable substituent (s), and
R⁶ is hydrogen, hydroxy or acyl (lower) alkylthio and a pharmaceutically acceptable salt thereof,
processes for their preparation and pharmaceutical compositions comprising them as an active ingredient.

The present invention relates to new polypeptide compound and a pharmaceutically acceptable salt thereof.

More particularly, it relates to new polypeptide compound and a pharmaceutically acceptable salt thereof, which have antimicrobial activities (especially, antifungal activities), inhibitory activity on β -1, 3-

5 glucan synthase, and further which are expected to be useful for the treatment or prevention of Pneumocystis carinii infection (e.g. Pneumocystis carinii pneumonia) in human being and animals, to a process for preparation thereof, to pharmaceutical composition comprising the same, and to a method for treating or preventing infectious diseases including Pneumocystis carinii infection (e.g. Pneumocystis carinii pneumonia) in human being and animals.

10 Accordingly, one object of the present invention is to provide the polypeptide compound and a pharmaceutically acceptable salt thereof, which are highly active against a number of pathogenic microorganisms and further which are expected to be useful for the treatment or prevention of Pneumocystis carinii infection (e.g. Pneumocystis carinii pneumonia) in human being or animals.

15 Another object of the present invention is to provide a process for the preparation of the polypeptide compound and a salt thereof.

A further object of the present invention is to provide a pharmaceutical composition comprising, as an active ingredient, said polypeptide compound or a pharmaceutically acceptable salt thereof.

20 Still further object of the present invention is to provide a method for treating or preventing infectious diseases including Pneumocystis carinii infection (e.g. Pneumocystis carinii pneumonia) caused by pathogenic microorganisms, which comprises administering said polypeptide compound to human being or animals.

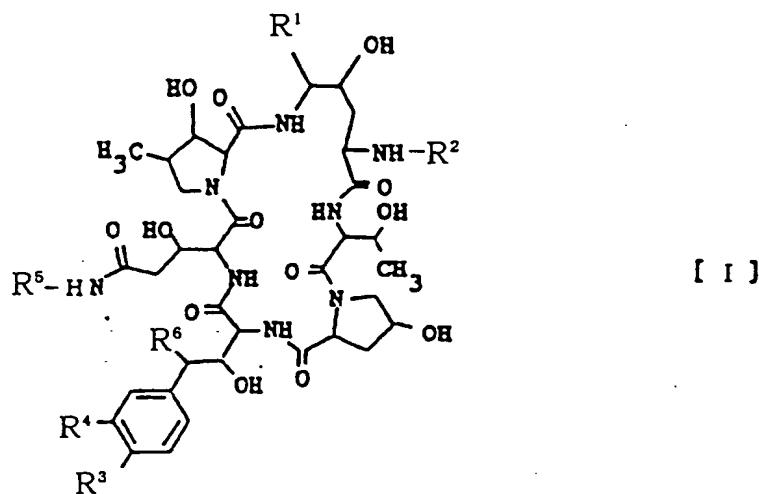
The object polypeptide compound of the present invention is novel and can be represented by the following general formula [I] :

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wherein

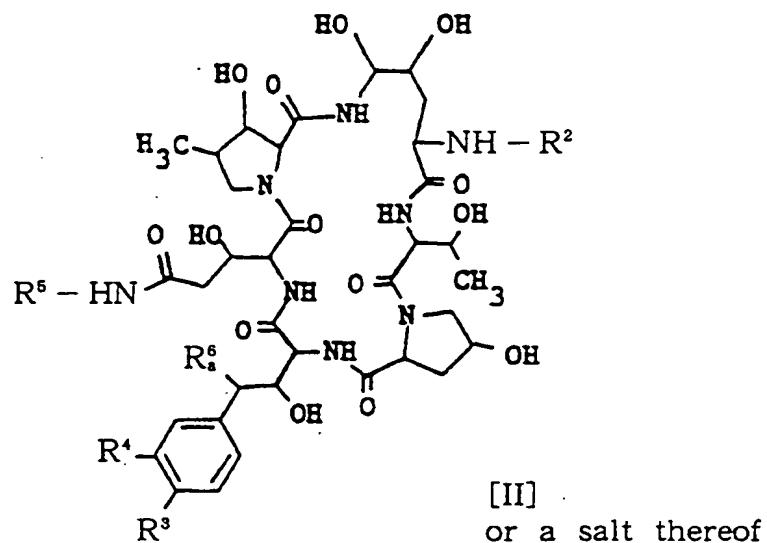
45 R¹ is hydrogen,
 R² is acyl group,
 R³ is hydroxy or acyloxy,
 R⁴ is hydroxy or hydroxysulfonyloxy,
 R⁵ is hydrogen or lower alkyl which may have one or more suitable substituent (s), and
 50 R⁶ is hydrogen, hydroxy, or acyl (lower) alkylthio.

The polypeptide compound [I] of the present invention can be prepared by the processes as illustrated in the following scheme.

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Process 1

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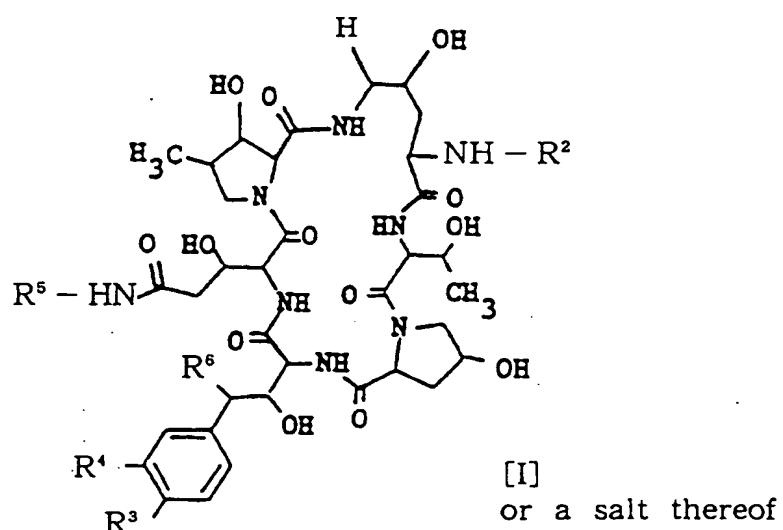
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Reduction
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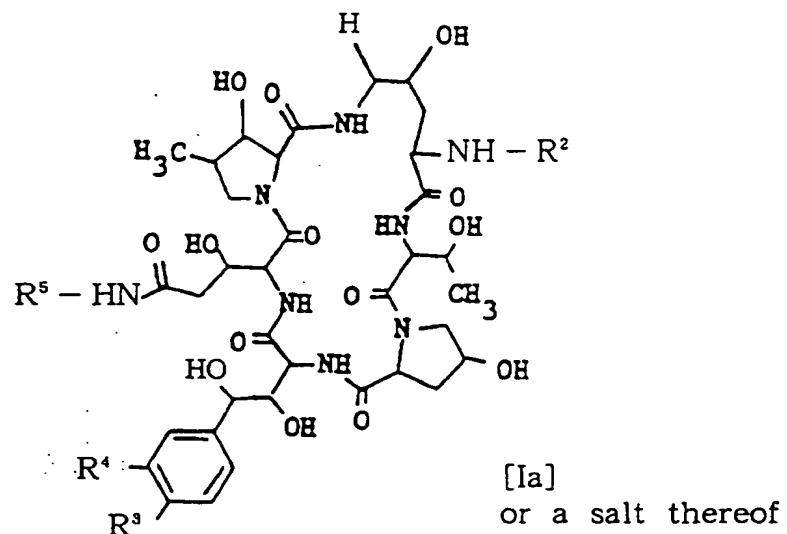
Process 2

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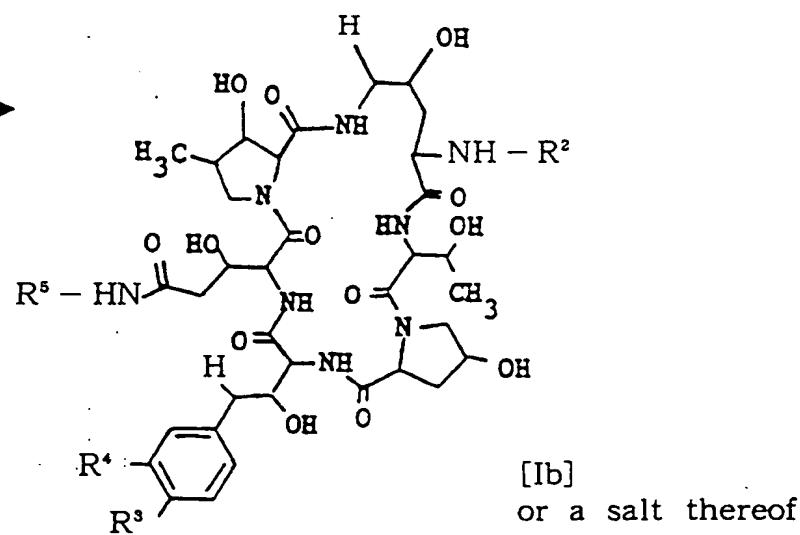
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Reduction
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Process 3

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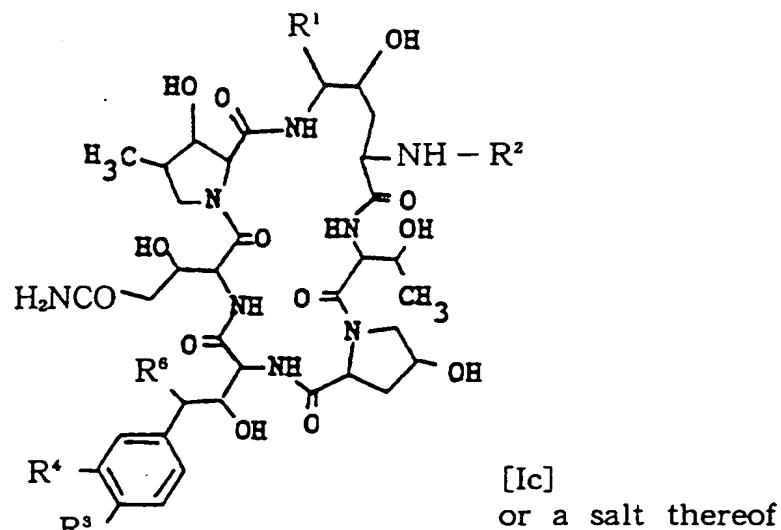
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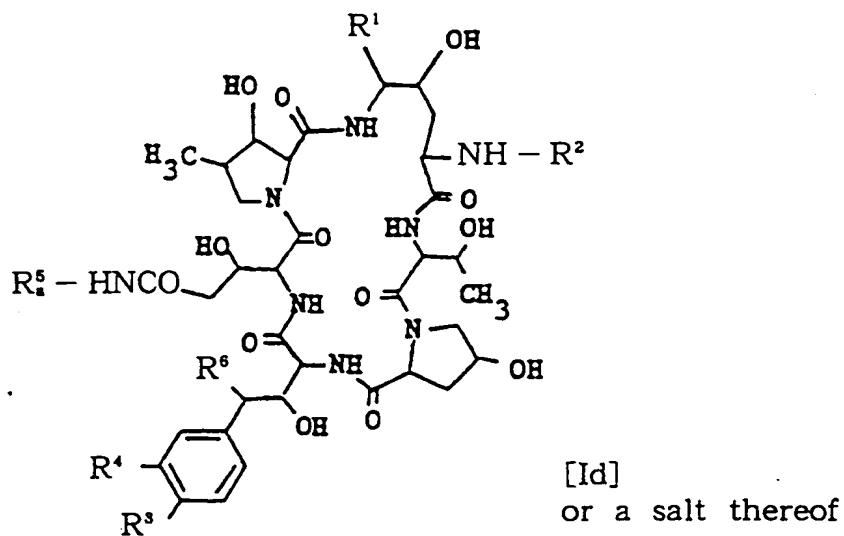
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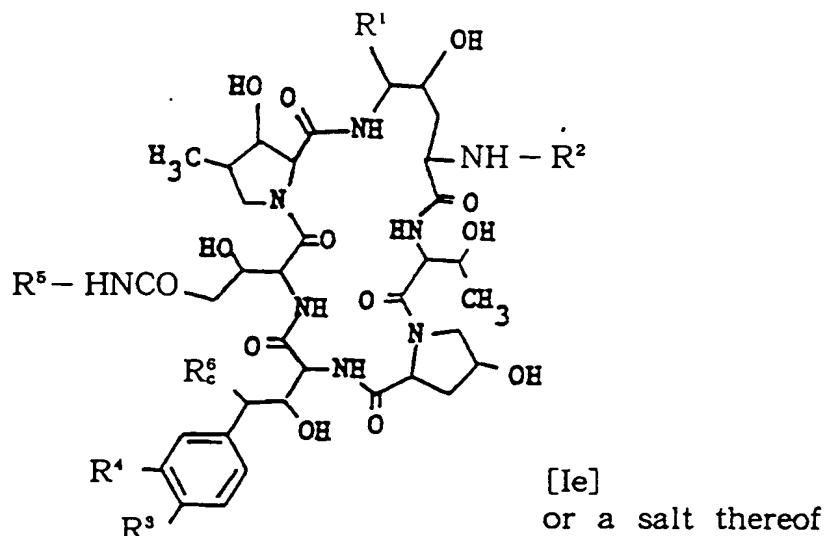


Alkylation



Process 4

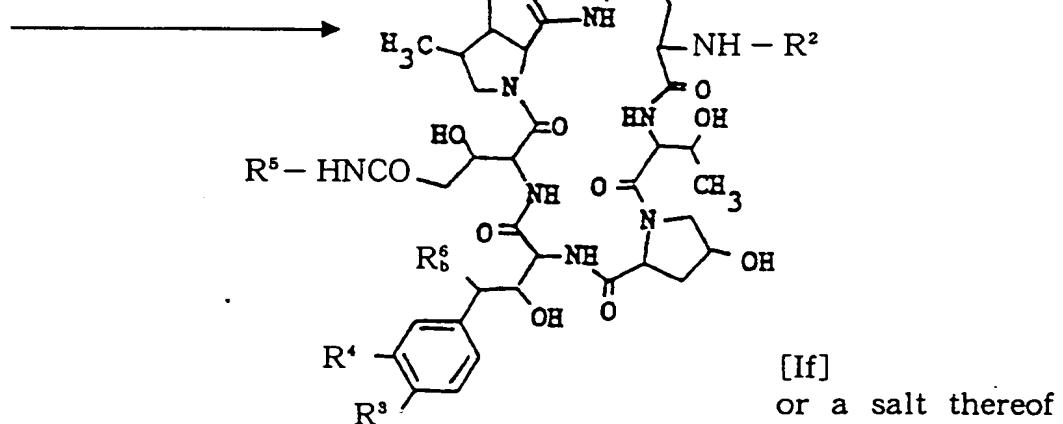
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Substitution reaction



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Process 5

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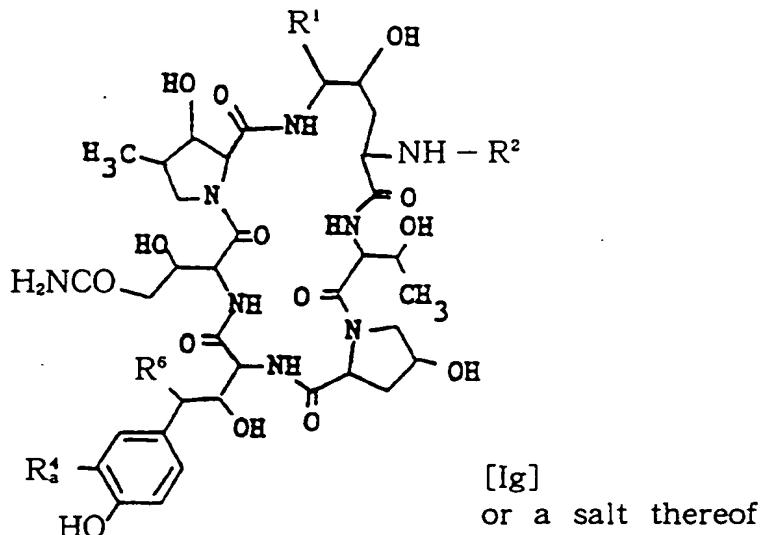
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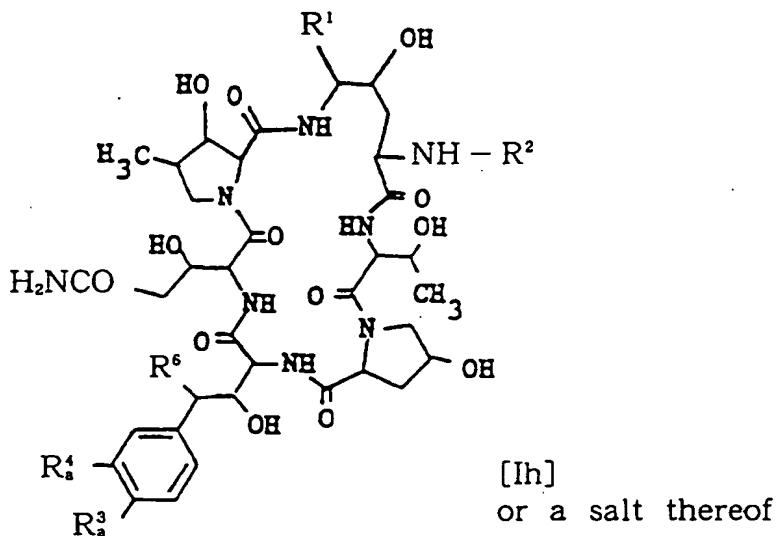
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Acylation

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Wherein

R¹, R², R³, R⁴, R⁵ and R⁶ are each as defined above,
 R³_a is acyloxy,
 R⁴_a is hydroxysulfonyloxy,
 R⁵_a is lower alkyl which may have one or more suitable substituent (s),
 R⁶_a is hydroxy, or acyl (lower) alkylthio,
 R⁶_b is acyl (lower) alkylthio,
 R⁶_c is hydroxy.

Suitable pharmaceutically acceptable salts of the object compound [I] is conventional non-toxic mono or di salts and include a metal salt such as an alkali metal salt [e.g. sodium salt, potassium salt, etc.] and an alkaline earth metal salt [e.g. calcium salt, magnesium salt, etc.], an ammonium salt, an organic base salt [e.g. trimethylamine salt, triethylamine salt, pyridine salt, picoline salt, dicyclohexylamine salt, N,N'-dibenzylethylenediamine salt, etc.], an organic acid addition salt [e.g. formate, acetate, trifluoroacetate, maleate, tartrate, methanesulfonate, benzenesulfonate, toluenesulfonate, etc.], an inorganic acid addition salt [e.g. hydrochloride, hydrobromide, hydriodide, sulfate, phosphate, etc.], a salt with an amino acid [e.g. arginine, salt, aspartic acid salt, glutamic acid salt, etc.], and the like.

In the above and subsequent description of this specification, suitable examples of the various definitions are explained in detail as follows :

The term "lower" is intended to mean 1 to 6 carbon atom (s), preferably 1 to 4 carbon atom (s), unless otherwise indicated.

5 The term "higher" is intended to mean 7 to 20 carbon atoms, preferably 7 to 15 carbon atoms, more preferably 7 to 10 carbon atoms unless otherwise indicated.

Suitable "acyl group" and "acyl" moiety in the term "acyloxy" may be aliphatic acyl, aromatic acyl, heterocyclic acyl, arylaliphatic acyl and heterocyclic-aliphatic acyl derived from carboxylic acid, carbonic acid, carbamic acid, sulfonic acid, and the like.

10 Suitable example of the "acyl" moiety thus explained may be : lower alkanoyl [e.g. formyl, acetyl, propionyl, butyryl, isobutyryl, valeryl, hexanoyl; pivaloyl, etc.] which may have one or more (preferably 1 to 3) suitable substituent (s) such as halogen [e.g. fluoro, chloro, bromo, iodo] ; aryl [e.g. phenyl, naphthyl, anthryl, etc.] which may have one or more (preferably 1 to 3) suitable substituent (s) like hydroxy, higher alkoxy as explained below, aforesaid aryl, or the like ; lower alkoxy as explained below ; amino ; protected 15 amino, preferably, acylamino such as lower alkoxy carbonylamino [e.g. methoxycarbonylamino, ethoxycarbonylamino, propoxycarbonylamino, butoxycarbonylamino, t-butoxycarbonylamino, pentyloxycarbonylamino, hexyloxycarbonylamino, etc.] ; or the like ; di (lower) alkylamino [e.g. dimethylamino, N-methylethylamino, diethylamino, N-propylbutylamino, dipentylamino, dihexylamino, etc.] ; lower alkoxyimino [e.g. methoxyimino, ethoxyimino, propoxyimino, butoxyimino, t-butoxyimino, pentyloxyimino, hexyloxyimino, etc.] ; ar 20 (lower) alkoxyimino such as phenyl (lower) alkoxyimino [e.g. benzyloxyimino, phenethyloxyimino, benzhydryloxyimino, etc.] which may have one or more (preferably 1 to 3) suitable substituent (s) like higher alkoxy as explained below, or the like ; heterocyclicthio, preferably, pyridylthio, which may have one or more (preferably 1 to 3) suitable substituent (s) like higher alkyl [e.g. heptyl, octyl, 2-ethylhexyl, nonyl, decyl, 3,7-dimethyloctyl, undecyl, dodecyl, tridecyl, tetradecyl, pentadecyl, 3-methyl-10-ethyldodecyl, hex- 25 adecyl, heptadecyl, octadecyl, nonadecyl, icosyl, etc.], or the like ; heteromonocyclic group [e.g. thienyl, imidazolyl, pyrazolyl, furyl; tetrazolyl, thiadiazolyl, etc.] which may have one or more (preferably 1 to 3) suitable substituent (s) like amino, aforesaid protected amino, aforesaid higher alkyl, or the like ; or the like ; higher alkanoyl [e.g. heptanoyl, octanoyl, nonanoyl, decanoyl, undecanoyl, lauroyl, tridecanoyl, myristoyl, pentadecanoyl, palmitoyl, 10,12-dimethyltetradecanoyl, heptadecanoyl, stearoyl, nonadecanoyl, 30 icosanoyl, etc.] ;

lower alkenoyl [e.g. acryloyl, methacryloyl, crotonoyl, 3-pentenoyl, 5-hexenoyl, etc.] which may have one or more (preferably 1 to 3) suitable substituent (s) such as aforesaid aryl which may have one or more (preferably 1 to 3) suitable substituent (s) like higher alkoxy as explained below, or the like, or the like ;

higher alkenoyl [e.g. 4-heptenoyl, 3-octenoyl, 3, 6-decadienoyl, 3, 7, 11-trimethyl-2, 6, 10-

35 dodecatrienoyl, 4, 10-heptadecadienoyl, etc.] ;

lower alkoxy carbonyl [e.g. methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, butoxycarbonyl, t-butoxycarbonyl, pentyloxycarbonyl, hexyloxycarbonyl, etc.] :

higher alkoxy carbonyl [e.g. heptyloxy carbonyl, octyloxy carbonyl, 2-ethylhexyloxy carbonyl, nonyloxy carbonyl, decyloxy carbonyl, 3,7-dimethyloctyloxy carbonyl, undecyloxy carbonyl, dodecyloxy carbonyl,

40 tridecyloxy carbonyl, tetradecyloxy carbonyl, pentadecyloxy carbonyl, 3-methyl-10-ethyldodecyloxy carbonyl, hexadecyloxy carbonyl, heptadecyloxy carbonyl, octadecyloxy carbonyl, nonadecyloxy carbonyl, icosyloxy carbonyl, etc.] ;

aryloxy carbonyl [e.g. phenoxy carbonyl, naphthylloxy carbonyl, etc.] ;

aryl glyoxyloyl [e.g. phenylglyoxyloyl, naphthylglyoxyloyl, etc.] ;

45 ar (lower) alkoxy carbonyl which may have one or more suitable substituent (s) such as phenyl (lower) alkoxy carbonyl which may have nitro or lower alkoxy [e.g. benzyloxy carbonyl, phenethyloxy carbonyl, p-nitrobenzylloxy carbonyl, p-methoxybenzylloxy carbonyl, etc.] ;

lower alkylsulfonyl [e.g. methylsulfonyl, ethylsulfonyl, propylsulfonyl, isopropylsulfonyl, pentylsulfonyl, butylsulfonyl, etc.] :

50 arylsulfonyl [e.g. phenylsulfonyl, naphthylsulfonyl, etc.] which may have one or more (preferably 1 to 3) suitable substituent (s) such as lower alkyl as explained below, higher alkoxy as explained below, or the like ;

ar (lower) alkylsulfonyl such as phenyl (lower) alkylsulfonyl [e.g. benzylsulfonyl, phenethylsulfonyl, benzhydrylsulfonyl, etc.], or the like.

55 aroyl [e.g. benzoyl, naphthoyl, anthrylcarbonyl, etc.] which may have one or more (preferably 1 to 5) suitable substituent (s) such as aforesaid halogen ; lower alkyl [e.g. methyl, ethyl, propyl, butyl, t-butyl, pentyl, hexyl, etc.] ; aforesaid higher alkyl ; lower alkoxy [e.g. methoxy, ethoxy, propoxy, butoxy, t-butoxy, pentyloxy, 4-methylpentyloxy, hexyloxy, etc.] which may have one or more (preferably 1 to 10) suitable

substituent (s) like aforesaid lower alkoxy, aforesaid halogen, aforesaid aryl, or the like ; higher alkoxy [e.g. heptyloxy, octyloxy, 2-ethylhexyloxy, nonyloxy, decyloxy, 3, 7-dimethyloctyloxy, undecyloxy, dodecyloxy, tridecyloxy, tetradecyloxy, pentadecyloxy, 3-methyl-10-ethylodecyloxy, hexadecyloxy, heptadecyloxy octadecyloxy, nonadecyloxy, icosyloxy, etc.] which may have one or more (preferably 1 to 17) suitable substituent (s) like aforesaid halogen ; higher alkenyloxy [e.g. 3-heptenyloxy, 7-octenyloxy, 2, 6-octadienyloxy, 5-nonenyloxy, 1-decenyloxy, 3, 7-dimethyl-6-octenyloxy, 3, 7-dimethyl-2, 6-octadienyloxy, 8-undecenyloxy, 3, 6, 8-dodecatrienyloxy, 5-tridecenyloxy, 7-tetradecenyloxy, 1, 8-pentadecadienyloxy, 15-hexadecenyloxy, 11-heptadecenyloxy, 7-octadecenyloxy, 10-nonadecenyloxy, 18-icosenyloxy, etc.] : carboxy ; aforesaid aryl which may have one or more (preferably 1 to 3) suitable substituent (s) like aforesaid higher alkoxy, aryloxy [e.g. phenoxy, naphthyoxy, anthryloxy, etc.] which may have one more (preferably 1 to 3) suitable substituent (s) like aforesaid higher alkoxy ; or the like ; or the like.

In the "acyl group", the preferred one may be aroyl which may have one or more (preferably 1 to 5) suitable substituent (s) such as i) lower alkoxy, ii) higher alkoxy, or iii) aryl which may have one or more (preferably 1 to 3) lower alkoxy or higher alkoxy ; in which the more preferred one may be benzoyl or naphthoyl, each of which may have 1 to 3 suitable substituent (s) selected from the group consisting of lower alkoxy, higher alkoxy and phenyl which may have 1 to 3 higher alkoxy, the much more preferred one may be benzoyl which may have 1 to 3 (C₇-C₁₆) alkoxy or phenyl which may have 1 to 3 (C₅-C₁₆) alkoxy ; or naphthoyl which may have 1 to 3 (C₇-C₁₆) alkoxy, and the most preferred one may be 4-octyloxybenzoyl, 4-(4-heptyloxyphenyl) benzoyl, 6-heptyloxy-2-naphthoyl and 6-octyloxy-2-naphthoyl, 4-(4-pentyloxyphenyl) benzoyl, 4-(4-nonyloxyphenyl) benzoyl, 4-(4-methylpentyloxyphenyl) benzoyl, 6-(4-methylpentyl) 2-naphthoyl, 6-hexyloxy-2-naphthoyl, 4-(4-hexyloxyphenyl) benzoyl ; and another preferred one may be aforesaid higher alkanoyl, in which the much more preferred one may be (C₇-C₁₇) alkanoyl, and the most preferred one may be palmitoyl.

The preferred "acyloxy" may be lower alkanoyloxy.

In the "acyl (lower) alkylthio", the preferred "acyl" moiety may be carboxy ; or protected carboxy such as esterified carboxy [e.g. lower alkoxy carbonyl (e.g. methoxycarbonyl, ethoxycarbonyl, propoxycarbon, butoxycarbonyl, t-butoxycarbonyl, pentyloxycarbonyl, hexyloxycarbonyl, etc) etc] ; in which the more preferred one may be carboxy.

Suitable "(lower) alkyl" moiety in the "acyl (lower) alkylthio" may be the ones as exemplified for "lower alkyl" in "lower alkyl which may have one or more suitable substituent (s)" before.

Suitable "lower alkyl" in "lower alkyl which may have one or more suitable substituent (s)" may include straight or branched ones having 1 to 6 carbon atom (s) such as methyl, ethyl, propyl, butyl, t-butyl, pentyl, hexyl, or the like, in which the preferred one may be (C₁-C₆) alkyl and the more preferred one may be methyl.

This "lower alkyl" may have one or more (preferably 1 to 3) suitable substituent (s) such as hydroxy, acyl [e.g. carboxy, protected carboxy {e.g. lower alkoxy carbonyl (e.g. methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, butoxycarbonyl, t-butoxycarbonyl, pentyloxycarbonyl, hexyloxycarbonyl, etc), etc}], di(lower) alkylamino [e.g. demethylamino, diethylamino, N-methylethylamino, dipropylamino, dibutylamino, N-t-butylbutylamino, dipentylamino, dehexylamino, etc], cyclic amino, in which it may have the other hetero atom (s) in its ring member [e.g. 1-pyrrolidinyl, piperidino, 1-piperazinyl, morpholino, 1-pyridyl, dihydropyridin-1-yl, etc], or the like.

The preferred example of "lower alkyl which has one or more suitable substituent (s)" may include lower alkyl having hydroxy and carboxy, lower alkyl having di(lower) alkylamino ; and lower alkyl having cyclic amino, in which the more preferred one may include (C₁-C₄) alkyl having hydroxy and carboxy ; (C₁-C₄) alkyl having di(C₁-C₄) alkylamino ; and (C₁-C₄) alkyl having piperidino, and the most preferred one may be 1-hydroxy-1-caboxymethyl, dimethylaminomethyl and piperidinomethyl.

In the compound (I) as explained above,

(1) the preferred one is the compound

wherein

R¹ is hydrogen,
 R² is aroyl which may have one or more suitable substituent (s) selected from the group consisting of i) lower alkoxy, ii) higher alkoxy and iii) aryl which may have one or more lower alkoxy or higher alkoxy ; or higher alkanoyl ;
 R³ is hydroxy or acyloxy,
 R⁴ is hydroxy or hydroxysulfonyloxy,
 R⁵ is hydrogen or lower alkyl which may have one or more suitable substituent (s) selected from the group consisting of hydroxy, acyl, di(lower) alkylamino and cyclic amino, and
 R⁶ is hydrogen, hydroxy or acyl (lower) alkylthio, and

(2) the more preferred one is the compound

wherein

5 R² is benzoyl or naphthoyl, each of which may have 1 to 3 suitable substituents (s) selected from the group consisting of lower alkoxy ; higher alkoxy ; and phenyl which may have 1 to 3 higher alkoxy,

R³ is hydroxy,

R⁵ is hydrogen, and

R⁶ is hydrogen or hydroxy,

R¹ and R⁴ are each as defined above, and

10 (3) the much more preferred one is the compound

wherein

R² is benzoyl having higher alkoxy, benzoyl having phenyl which has higher alkoxy, naphthoyl having lower alkoxy, or naphthoyl having higher alkoxy, and

15 R⁶ is hydrogen,

R¹, R³, R⁴ and R⁵ are each as defined in (2) and

(4) the still more preferred one may be

i) the compound wherein R² is benzoyl having higher alkoxy,

R¹, R³, R⁴, R⁵ and R⁶ are each as defined in (3), or

20 ii) the compound wherein R² is benzoyl having phenyl which has higher alkoxy,

R¹, R³, R⁴, R⁵ and R⁶ are each as defined in (3), or

iii) the compound wherein R² is naphthoyl having lower alkoxy,

R¹, R³, R⁴, R⁵ and R⁶ are each as defined in (3), or

iv) the compound wherein R² is naphthoyl having higher alkoxy,

25 R¹, R³, R⁴, R⁵ and R⁶ are each as defined in (3), and

(5) another much more preferred one is the compound

wherein

R¹ is hydrogen,

R² is benzoyl having higher alkoxy,

30 R³ is benzoyl having phenyl which has higher alkoxy, naphthoyl having lower alkoxy, or naphthoyl having higher alkoxy,

R⁴ is hydroxy,

R⁵ is hydroxy or hydroxysulfonyloxy,

35 R⁶ is hydrogen, and

R⁶ is hydroxy, and

(6) another still more preferred one may be

i) the compound wherein R² is benzoyl having higher alkoxy,

R¹, R³, R⁴, R⁵ and R⁶ are each as defined in (5), or

40 ii) the compound wherein R² is benzoyl having phenyl which has higher alkoxy,

R¹, R³, R⁴, R⁵ and R⁶ are each as defined in (5), or

iii) the compound wherein R² is naphthoyl having lower alkoxy,

R¹, R³, R⁴, R⁵ and R⁶ are each as defined in (5), or

iv) the compound wherein R² is naphthoyl having higher alkoxy,

R¹, R³, R⁴, R⁵ and R⁶ are each as defined in (5).

45 The processes for preparing the object compound [I] or a salt thereof of the present invention are explained in detail in the following.

Process 1

50 The object compound [I] or its salt can be prepared by reducing compound [II] or its salt.

Suitable salts of the compounds [I] and [II] may be the same as those exemplified for the compound [I].

The reaction can be carried out in a conventional manner, namely, chemical reduction or catalytic reduction.

55 Suitable reducing agents to be used in chemical reduction are a combination of metal [e.g. tin, zinc, iron, etc.] or metallic compound [e.g. chromium chloride, chromium acetate, etc.] and an organic or inorganic acid [e.g. formic acid, acetic acid, propionic acid, trifluoroacetic acid, p-toluenesulfonic acid, hydrochloric acid, hydrobromic acid, hydride transfer reagent such as aluminum hydride compound (e.g.

lithium aluminum hydride, lithium hydridotri-t-butoxyaluminate, etc.), borohydride compound (e.g. sodium borohydride, sodium cyanoborohydride, etc.) or the like etc.].

Suitable catalysts to be used in catalytic reduction are conventional ones such as platinum catalyst [e.g. platinum plate, spongy platinum, platinum black, colloidal platinum, platinum oxide, platinum wire, etc.], palladium catalyst [e.g. spongy palladium, palladium black, palladium oxide, palladium on carbon, colloidal palladium, palladium on barium sulfate, palladium on barium carbonate, etc.], nickel catalyst [e.g. reduced nickel, nickel oxide, Raney nickel, etc.], cobalt catalyst [e.g. reduced cobalt, Raney cobalt, etc.], iron catalyst [e.g. reduced iron, Raney iron, etc.], copper catalyst [e.g. reduced copper, Raney copper, Ullman copper, etc.] or the like.

10 The reaction of this process is usually carried out in a solvent such as water, alcohol [e.g. methanol, ethanol, propanol, etc.], acetic acid, diethyl ether, dioxane, tetrahydrofuran, methylene chloride, etc. or a mixture thereof.

The reaction is preferably carried out under somewhat milder conditions such as under cooling to warming.

15 It is included within the scope of the present invention that "hydroxy" in R_a^6 may be reduced to "hydrogen" during the reaction.

Process 2

20 The object compound [lb] or a salt thereof can be prepared by reducing the compound [la] or a salt thereof to reduction reaction.

This reaction can be carried out in substantially the same manner as Process 1, and therefore the reaction mode and reaction conditions [e.g. solvent, reaction temperature, etc.] of this reaction are to be referred to those as explained in Process 1.

25 Process 3

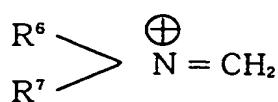
The object compound [ld] or a salt thereof can be prepared by subjecting the compound [lc] or a salt thereof to alkylation reaction.

30 Suitable agent for this alkylation reaction may include

1) lower alkanal [e.g. formaldehyde, ethanal, propanal, butanal, t-butanal, pentanal, hexanal, etc.] which may have one or more (preferably 1 to 3) suitable substituent (s) as exemplified for the substituent (s) of "lower alkyl" (which may have one or more suitable substituent (s)).

2) a compound of the formula :

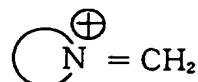
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(wherein R^6 and R^7 are each lower alkyl as exemplified before),

3) a compound of the formula :

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50 [wherein a group of the formula :

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is cyclic aminium group, in which it may have the other hetero atom (s) in its ring member (e.g. pyrrolidinium, piperidinium, piperadinium, morpholinium, etc)], and the like.

This reaction is usually carried out in a conventional solvent which does not adversely influence the reaction such as water, alcohol [e.g. methanol, ethanol, propanol, etc], tetrahydrofuran, dioxane, dimethyl sulfoxide, N,N-dimethylformamide, acetone, or a mixture thereof.

5 The reaction temperature is not critical, and the reaction is usually carried out under cooling to warming.

Process 4

10 The object compound [If] or a salt thereof can be prepared by subjecting the compound [Ie] or a salt thereof to substitution reaction with acyl (lower) alkylthiol.

This reaction is usually carried out in a solvent such as water, alcohol (e.g., methanol, ethanol, etc.), benzene, N, N-dimethylformamide, tetrahydrofuran, toluene, methylene chloride, ethylene dichloride, chloroform, dioxane, diethyl ether or any other solvents which do not adversely affect the reaction, or the mixture thereof.

15 The reaction temperature is not critical and the reaction is usually carried out under cooling to heating. The reaction is usually carried out in the presence of an acid including Lewis acid.

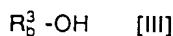
Suitable acid may include an organic [e.g. formic acid, acetic acid, propionic acid, trichloroacetic acid, trifluoroacetic acid, etc.] and an inorganic acid [e.g. hydrochloric acid, hydrobromic acid, sulfuric acid, hydrogen chloride, hydrogen bromide, zinc halide (e.g. zinc chloride, zinc bromide, etc.), etc.] and the like.

20 When the acid and/or the starting compound are in liquid, they can be used also as a solvent.

Process 5

25 The object compound [Ih] or a salt thereof can be prepared by subjecting the compound [Ig] or a salt thereof to acylation reaction.

Suitable acylating agent to be used in the present acylation reaction may include the compound of the formula :



30

[wherein R_b^3 is lower alkanoyl]

or its reactive derivative or a salt thereof.

Suitable "lower alkanoyl" in the compound [III] may include the ones as exemplified for "lower alkanoyl" moiety in "lower alkanoyloxy" before.

35 Suitable reactive derivative at the carboxy group of the compound [III] may include an acid halide, an acid anhydride, an activated amide, an activated ester, and the like. Suitable examples of the reactive derivatives may be an acid chloride; an acid azide; a mixed acid anhydride with an acid such as substituted phosphoric acid [e.g. dialkylphosphoric acid, phenylphosphoric acid, diphenylphosphoric acid, dibenzylphosphoric acid, halogenated phosphoric acid, etc.], dialkylphosphorous acid, sulfurous acid, thiosulfuric acid, sulfuric acid, sulfonic acid [e.g. methanesulfonic acid, etc.], aliphatic carboxylic acid [e.g. acetic acid, propionic acid, butyric acid, isobutyric acid, pivaric acid, pentanoic acid, isopentanoic acid, 2-ethylbutyric acid, trichloroacetic acid, etc.]; or aromatic carboxylic acid [e.g. benzoic acid, etc.]; a symmetrical acid anhydride [e.g. acetic anhydride, etc.]; an activated amide with imidazole, 4-substituted imidazole, dimethylpyrazole, triazole, tetrazole or 1-hydroxy-1H-benzotriazole; or an activated ester [e.g. cyanomethyl ester, methoxymethyl ester, dimethyliminomethyl $[(CH_3)_2N^+ = CH^-]$ ester, vinyl ester, propargyl ester, p-nitrophenyl ester, 2,4-dinitrophenyl ester, trichlorophenyl ester, pentachlorophenyl ester, mesylphenyl ester, phenylazophenyl ester, phenyl thioester, p-nitrophenyl thioester, p-cresyl thioester, carboxymethyl thioester, pyranyl ester, pyridyl ester, piperidyl ester, 8-quinolyl thioester, etc.], or an ester with a N-hydroxy compound [e.g. N,N-dimethylhydroxylamine, 1-hydroxy-2-[1H]-pyridone, N-hydroxysuccinimide, N-hydroxyphthalimide, 1-hydroxy-1H-benzotriazole, etc.], and the like. These reactive derivatives can optionally be selected from them according to the kind of the compound [III] to be used.

Suitable salts of the compound [III] and its reactive derivative can be referred to the ones as exemplified for the compound [I].

55 The reaction is usually carried out in a conventional solvent such as water, alcohol [e.g. methanol, ethanol, etc.], acetone, dioxane, acetonitrile, chloroform, methylene chloride, ethylene chloride, tetrahydrofuran, ethyl acetate, N,N-dimethylformamide, pyridine or any other organic solvent which does not adversely influence the reaction. These conventional solvent may also be used in a mixture with water.

In this reaction, when the compound [III] is used in a free acid form or its salt form, the reaction is preferably carried out in the presence of a conventional condensing agent such as N,N'-dicyclohexylcarbodiimide ; N-cyclohexyl-N'-morpholinoethylcarbodiimide ; N-cyclohexyl-N'-(4-diethylaminocyclohexyl) carbodiimide ; N,N'-diethylcarbodiimide, N,N'-diisopropylcarbodiimide ; N-ethyl-N'-(3-dimethylaminopropyl) carbodiimide ; N,N'-carbonylbis-(2-methylimidazole) ; pentamethyleneketene-N-cyclohexylimine ; diphenylketene-N-cyclohexylimine ; ethoxyacetylene ; 1-alkoxy-1-chloroethylene ; trialkyl phosphite ; ethyl polyphosphate ; isopropyl polyphosphate ; phosphorus oxychloride (phosphoryl chloride) ; phosphorus trichloride ; thionyl chloride ; oxalyl chloride ; lower alkyl haloformate [e.g. ethyl chloroformate, isopropyl chloroformate, etc.] ; triphenylphosphine ; 2-ethyl-7-hydroxybenzisoxazolium salt ; 2-ethyl-5-(m-sulfophenyl) isoxazolium hydroxide intramolecular salt ; 1-(p-chlorobenzenesulfonyloxy)-6-chloro-1H-benzotriazole ; so-called Vilsmeier reagent prepared by the reaction of N,N-dimethylformamide with thionyl chloride, phosphogene, trichloromethyl chloroformate, phosphorus oxychloride, methanesulfonyl chloride, etc. ; or the like.

The reaction may also be carried out in the presence of an inorganic or organic base such as an alkali metal carbonate, alkali metal bicarbonate, tri (lower) alkylamine [e.g. triethylamine, etc.], pyridine, di (lower) alkylaminopyridine [e.g. 4-dimethylaminopyridine, etc.], N-(lower) alkylmorpholine, N,N-di (lower) alkylbenzylamine, or the like.

The reaction temperature is not critical, and the reaction is usually carried out under cooling to warming.

The starting compounds [II] can be prepared by the fermentation and synthetic processes disclosed in EP 0462531 A2 and the Preparations in this specification.

A culture of Coleophoma sp. F-11899, which is used in said fermentation process, has been deposited with the Fermentation Research Institute Agency of Industrial Science and Technology (1-3, Higashi 1 chome, Tsukuba-shi, IBARAKI 305 JAPAN) on October 26, 1989 under the number of FERM BP-2635.

25 Biological property of the polypeptide compound [I] of the present invention

In order to show the usefulness of the polypeptide compound [I] of the present invention, the biological data of the representative compound is explained in the following.

30 Test A (Antimicrobial activity) :

In vitro antimicrobial activity of the compound of Example 19 (Major Compounds) disclosed later were determined by the two -fold agar-plate dilution method as described below.

35 Test Method

One loopful of an overnight culture of each test microorganism in Sabouraud broth containing 2% Glucose (10^5 viable cells per ml) was streaked on yeast nitrogen base dextrose agar (YNBDA) containing graded concentrations of the compound [I], and the minimal inhibitory concentration (MIC) was expressed in terms of μ g/ml after incubation at 30 °C for 24 hours.

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Test ResultMIC (μ g/ml)

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Test Compound	The compound of <u>Example 19</u> (Major Compound)
Test organism	
Candida albicans YU - 1200	0.05

From the test result, it is realized that the polypeptide compound [I] of the present invention has an antimicrobial activity (especially, antifungal activity).

The pharmaceutical composition of this invention can be used in the form of a pharmaceutical preparation, for example, in solid, semisolid or liquid form, which contains the polypeptide compound [I] or a pharmaceutically acceptable salt thereof, as an active ingredient in admixture with an organic or inorganic carrier, or excipient suitable for rectal, pulmonary (nasal or buccal inhalation), nasal, ocular, external (topical), oral, or parenteral (including subcutaneous, intravenous and intramuscular) administration or insufflation. The active ingredient may be compound, for example, with the usual non-toxic, pharmaceutically acceptable carriers for tablets, pellets, troches, capsules, suppositories, creams, ointments, aerosols, powders for insufflation, solutions, emulsions, suspensions, and any other form suitable for use. And, If necessary, in addition, auxiliary, stabilizing, thickening and coloring agents and perfumes may be used. The polypeptide compound [I] or a pharmaceutically acceptable salt thereof is/are included in the pharmaceutical composition in an amount sufficient to produce the desired antimicrobial effect upon the process or condition of diseases. For applying the composition to human, it is preferable to apply it by intravenous, intramuscular, pulmonary, or oral administration, or insufflation. While the dosage of therapeutically effective amount of the polypeptide compound [I] varies from and also depends upon the age and condition of each individual patient to be treated, in the case of intravenous administration, a daily dose of 0.01-20 mg of the polypeptide compound [I] per kg weight of human being in the case of intramuscular administration, a daily dose of 0.1-20 mg of the polypeptide compound [I] per kg weight of human being, in case of oral administration, a daily dose of 0.5-50 mg of the polypeptide compound [I] per kg weight of human being is generally given for treating or preventing infectious diseases.

Especially in case of the treatment or prevention of Pneumocystis carinii infection, the followings are to be noted.

For administration by inhalation, the compounds of the present invention are conveniently delivered in the form of an aerosol spray presentation from pressurized as powders which may be formulated and the powder compositions may be inhaled with the aid of an insufflation powder inhaler device. The preferred delivery system for inhalation is a metered dose inhalation aerosol, which may be formulated as a suspension or solution of compound in suitable propellants such as fluorocarbons or hydrocarbons.

Because of desirability to directly treat lung and bronchi, aerosol administration is a preferred method of administration. Insufflation is also a desirable method, especially where infection may have spread to ears and other body cavities.

Alternatively, parenteral administration may be employed using drip intravenous administration.

The following Preparations and Examples are given for the purpose of illustrating the present invention in more detail.

Preparation 1

To a solution of methyltrichlorosilane (0.82ml) in acetonitrile was added 1-ethoxymethylpiperidine (1g) under ice-cooling and stirred for five minutes at the same temperature. Diethyl ether was added thereto and the produced precipitate was collected by filtration and dried to give 1-methylenepiperidinium chloride (0.88g).

IR (Nujol) : 3100, 2700, 1420, 1110, 920 cm^{-1}

EI-MS : $e/z = 133 (M +)$ Preparation 2

5 A solution of 6-hydroxy-2-naphthoic acid (1.04g) in a mixture of 10% sodium hydroxide aqueous solution (4.44ml) and dimethyl sulfoxide (18ml) was stirred for half an hour at 80 °C. Then heptyl bromide (0.872ml) was added thereto and stirred for 5 hours at 60 °C. The reaction mixture was added to water (50ml) and the mixture was adjusted to pH 3 with conc. hydrochloric acid. The resultant precipitate was collected by filtration and dried to give 6-heptyloxy-2-naphthoic acid (1.39g).

10 IR (Nujol) : 1660, 1620, 1210 cm^{-1}

NMR (DMSO-d₆, δ) : 0.88 (3H, t, $J = 6.6\text{Hz}$), 1.2-1.6 (8H, m), 1.7-1.9 (2H, m), 4.10 (2H, t, $J = 6.5\text{Hz}$), 7.18 (1H, dd, $J = 8.9\text{Hz}$ and 2.4Hz), 7.35 (1H, d, $J = 2.4\text{Hz}$), 7.79 (1H, d, $J = 8.6\text{Hz}$), 7.9-8.1 (2H, m), 8.45 (1H, s)

Preparation 3

The following compound was obtained according to a similar manner to that of Preparation 2.

4-(4-pentyloxyphenyl)benzoic acid

IR(KBr) : 1678, 1605, 1200, 833 cm^{-1}

Preparation 4

The following compound was obtained according to a similar manner to that of Preparation 2.

4-(4-nonyloxyphenyl)benzoic acid

IR(Nujol) : 1686, 1604, 1203, 837 cm^{-1}

Preparation 5

The following compound was obtained according to a similar manner to that of Preparation 2.

30 6-(4-methylpentyloxy)-2-naphthoic acid

IR(KBr) : 1674, 1624, 1292, 1213 cm^{-1}

NMR (CDCl₃, δ) : 0.64(6H, d, $J = 6.6\text{Hz}$), 1.3-1.5(2H, m), 1.65(1H, m), 1.78-2.0(2H, m), 4.08(2H, t, $J = 6.6\text{Hz}$), 7.15(1H, d, $J = 2.3\text{Hz}$), 7.21(1H, dd, $J = 2.3$ and 8.9Hz), 7.76(1H, d, $J = 8.7\text{Hz}$), 7.87(1H, d, $J = 8.9\text{Hz}$), 8.08(1H, dd, $J = 2.3$ and 8.7Hz), 8.63(1H, s)

Preparation 6

To a susupension of N-hydroxysuccinimide (0.56g) and 6-heptyloxy-2-naphthoic acid (1.39g) in methylene chloride (42ml) was added 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride (1.21g) and stirred for 3 hours at room temparature. The reaction mixture was added to water (100ml). The organic layer was separated and dried over magnesium sulfate. Magnesium sulfate was filtered off, and the filtrate was evaporated under reduced pressure to give succinimido 6-heptyloxy-2-naphthoate (1.48g).

IR (Nujol) : 1760, 1740, 1620 cm^{-1}

NMR (CDCl₃, δ) : 0.91 (3H, t, $J = 6.6\text{Hz}$), 1.2-1.6 (8H, m), 1.7-2.0 (2H, m), 2.93 (4H, s), 4.10 (2H, t, $J = 6.5\text{Hz}$), 7.1-7.3 (2H, m), 7.77 (1H, d, $J = 8.6\text{Hz}$), 7.84 (1H, d, $J = 8.6\text{Hz}$), 8.03 (1H, dd, $J = 8.9\text{Hz}$ and 2.4Hz), 8.65 (1H, s)

Preparation 7

50 The following compound was obtained according to a similar manner to that of Preparation 6.

1-[4-(4-pentyloxyphenyl)benzoyl]1H-benzotriazole-3-oxide

IR(KBr) : 1776, 1605, 1194, 985 cm^{-1}

NMR (CDCl₃, δ) : 0.96(3H, t, $J = 7.0\text{Hz}$), 1.3-1.6(4H, m), 1.84(2H, quint, $J = 6.8\text{Hz}$), 4.03(2H, t, $J = 6.5\text{Hz}$), 7.03(2H, d, $J = 8.7\text{Hz}$), 7.35-7.68(3H, m), 7.63(2H, d, $J = 8.7\text{Hz}$), 7.79(2H d, $J = 8.4\text{Hz}$), 8.12(1H, d, $J = 8.2\text{Hz}$), 8.32(2H, d, $J = 8.4\text{Hz}$)

Preparation 8

The following compound was obtained according to a similar manner to that of Preparation 6.

1-[4-(4-nonyloxyphenyl) benzoyl]-1H-benzotriazole-3-oxide

5 IR(KBr) : 1774, 1600 cm^{-1}

NMR (CDCl_3 , δ) : 0.89 (3H,t,J = 6.8Hz), 1.1-1.6 (12H,m), 1.83 (2H,quint, J = 6.5Hz), 4.03 (2H,t,J = 6.5Hz), 7.03 (2H,d,J = 8.8Hz), 7.35-7.68 (3H, m), 7.64 (2H,d,J = 6.8Hz), 7.79 (2H,d,J = 6.8Hz), 8.12 (1H,d,J = 9.2Hz), 8.32 (2H,d,J = 6.8Hz)

10 Preparation 9

The following compound was obtained according to a similar manner to that of Preparation 6.

1-[6-(4-methylpentyloxy)-2-naphthoyl]-1H-benzotriazole -3-oxide

IR(KBr) : 1784, 1628, 1196 cm^{-1}

15 The Starting Compounds used and the Object Compounds obtained in the following Preparations and Examples are given in the table as below, in which the formula of the starting compounds are in the upper column and the formula of the object compounds are in the lower column, respectively.

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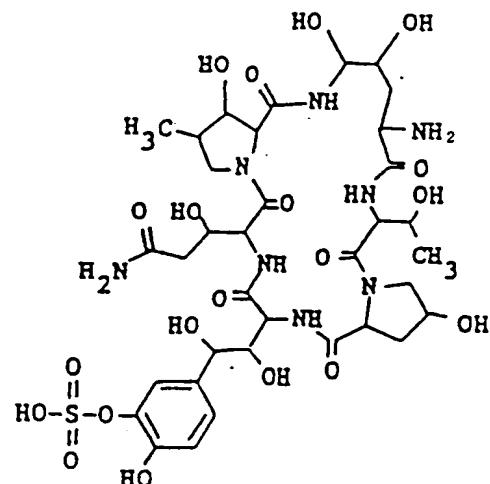
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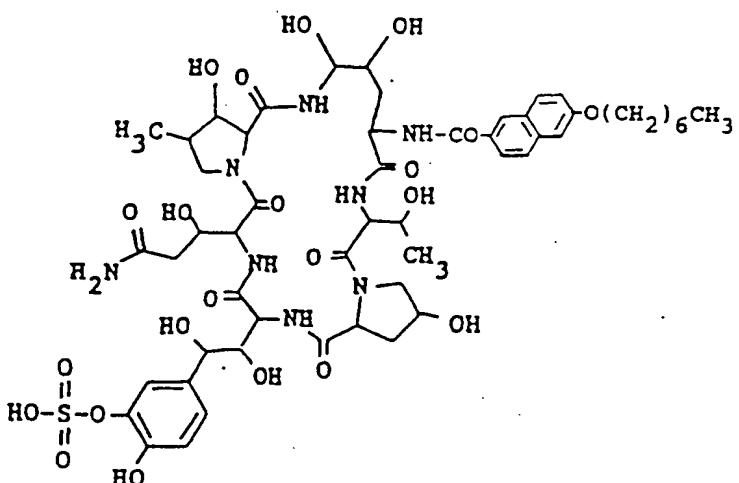
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Preparation
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Formula



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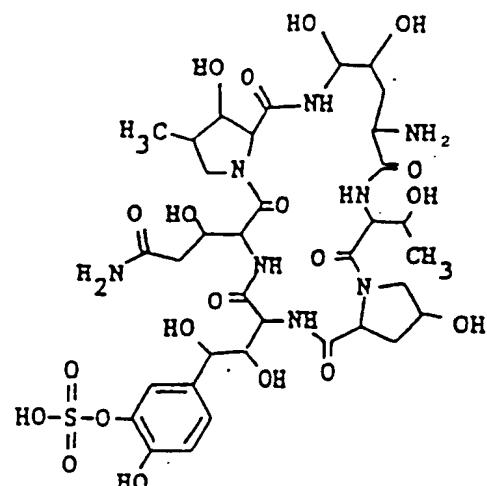
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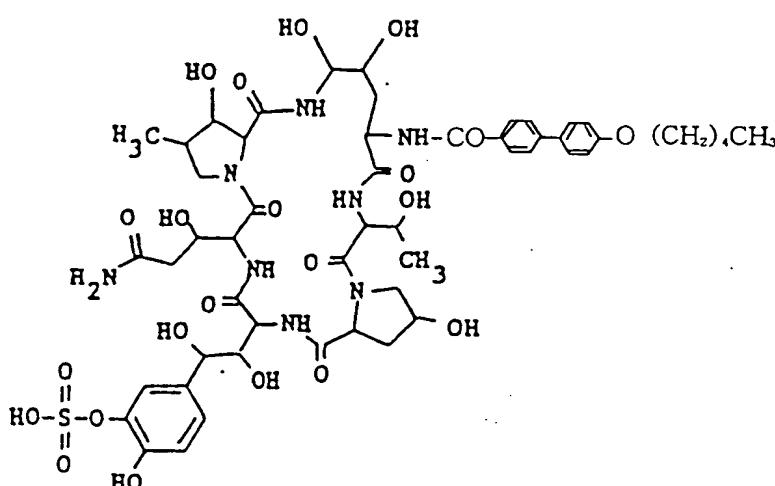
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Preparation
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Preparation No.	Formula
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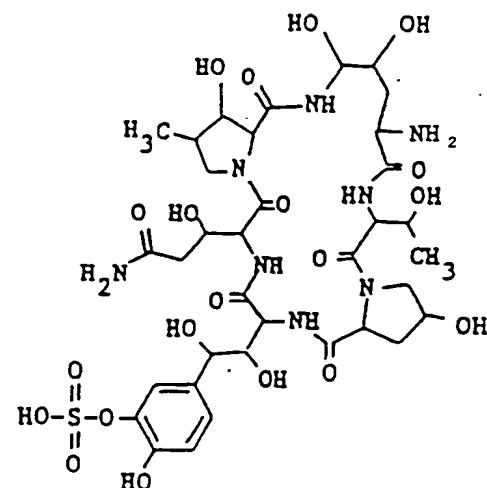
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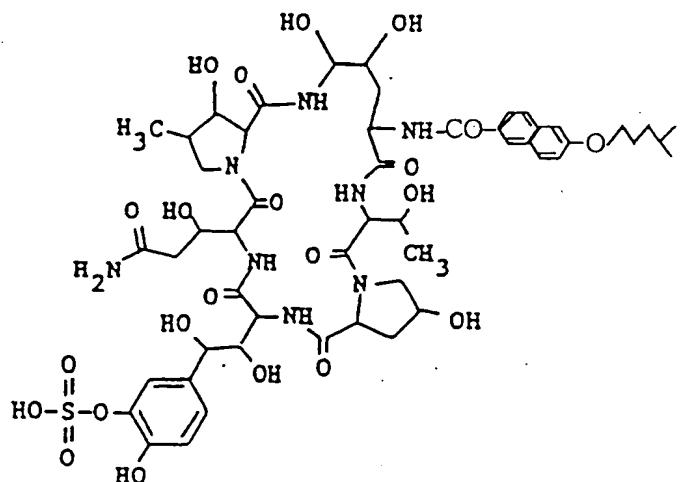
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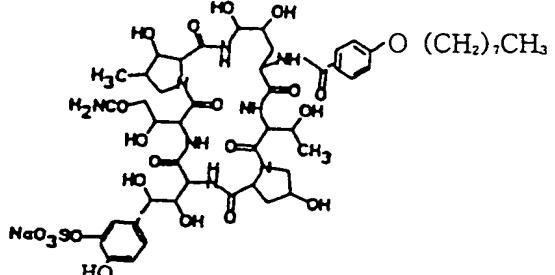
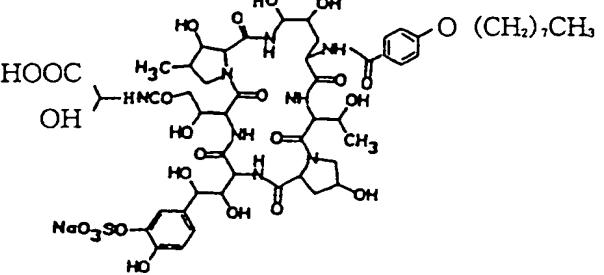
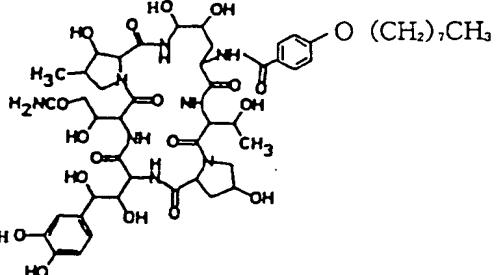
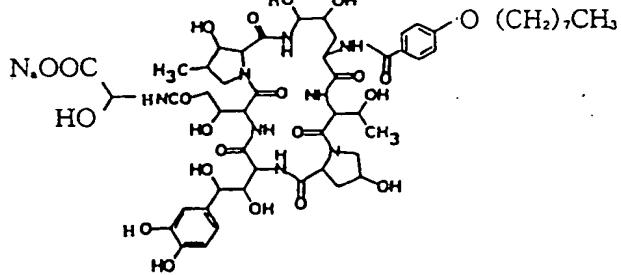
Formula



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Preparation No.	F o r m u l a
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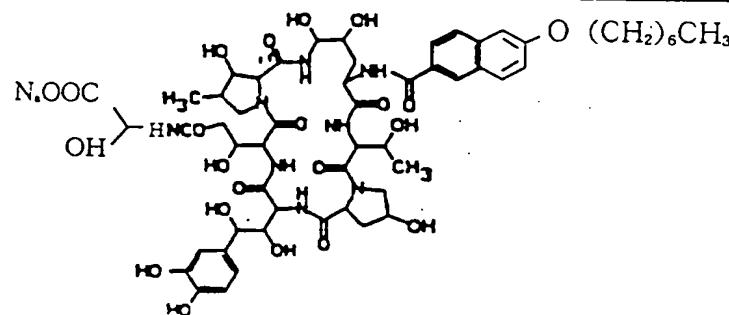
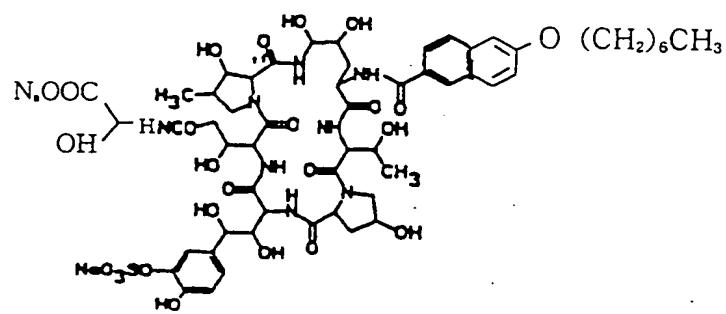
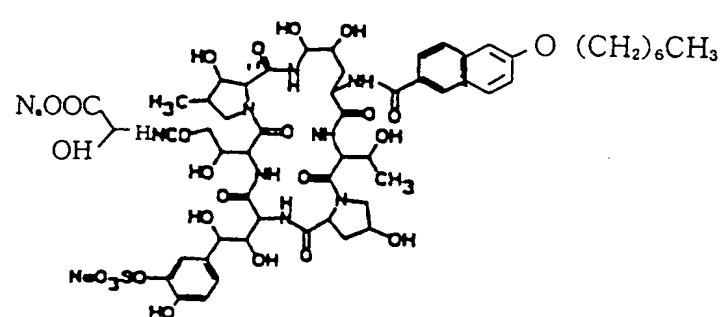
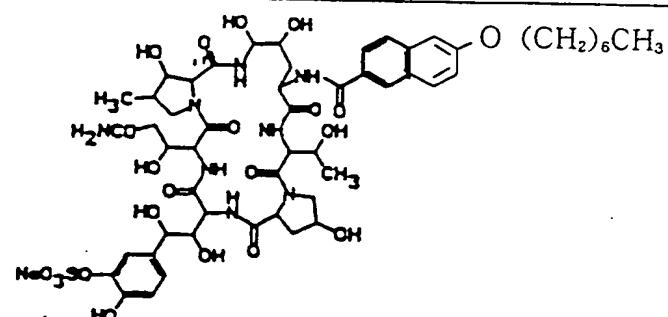
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Preparation
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Formula



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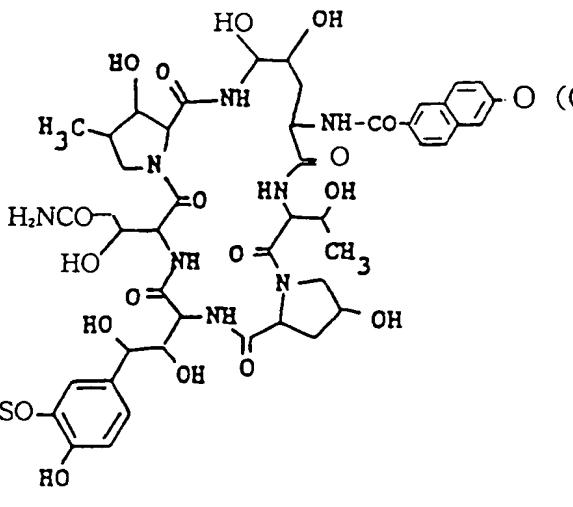
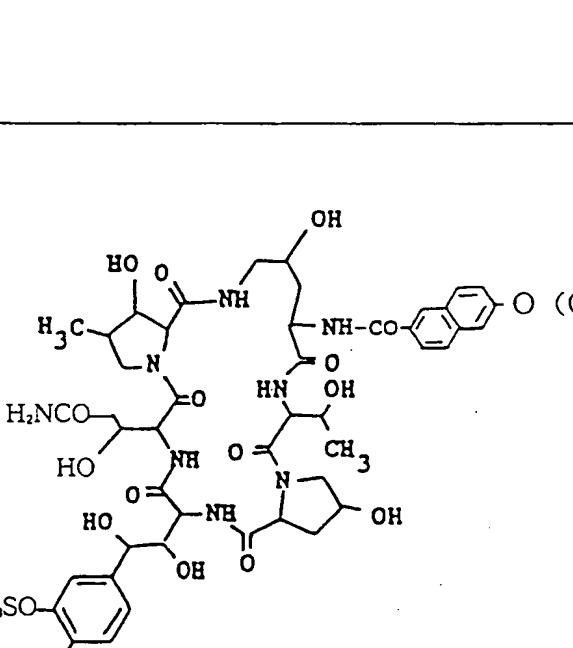
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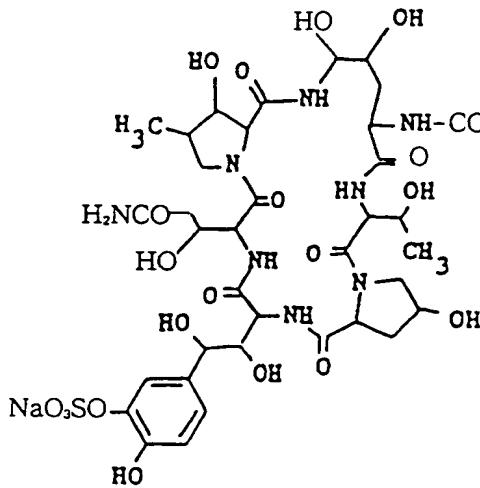
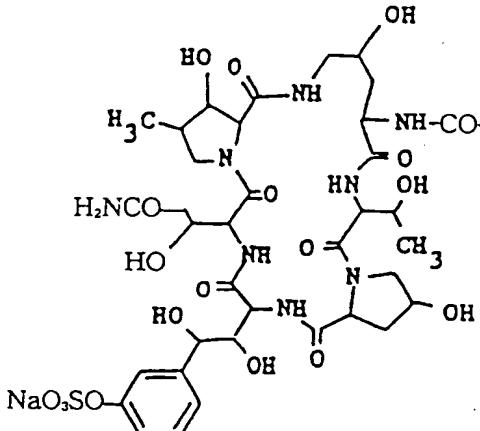
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Example No.	Formula
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2 30 35 40 45 50	

Example No.	Formula
	
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Example No.	Formula
	
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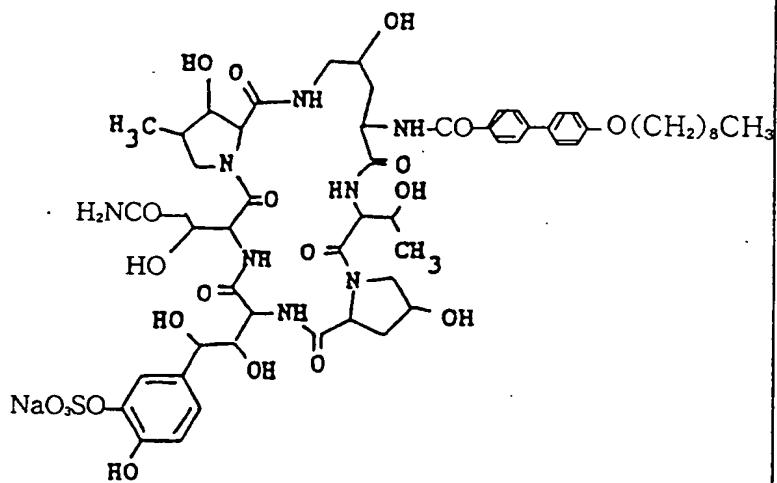
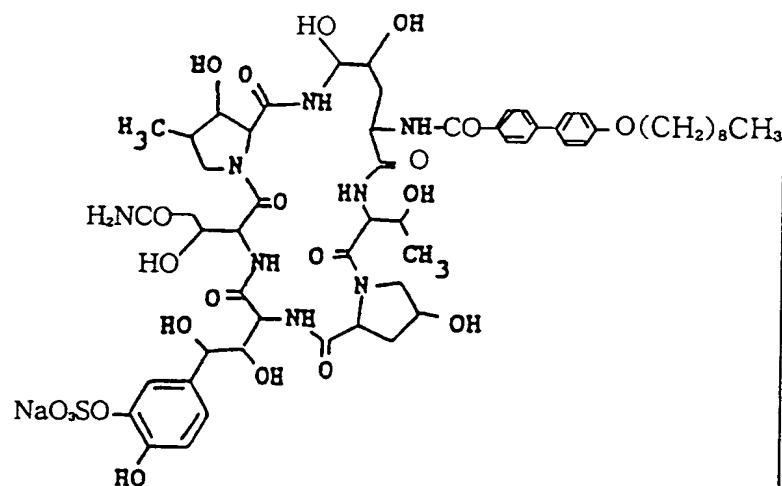
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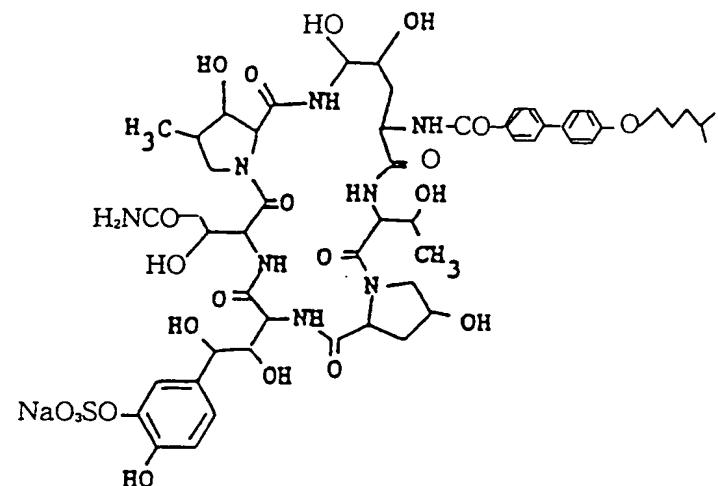
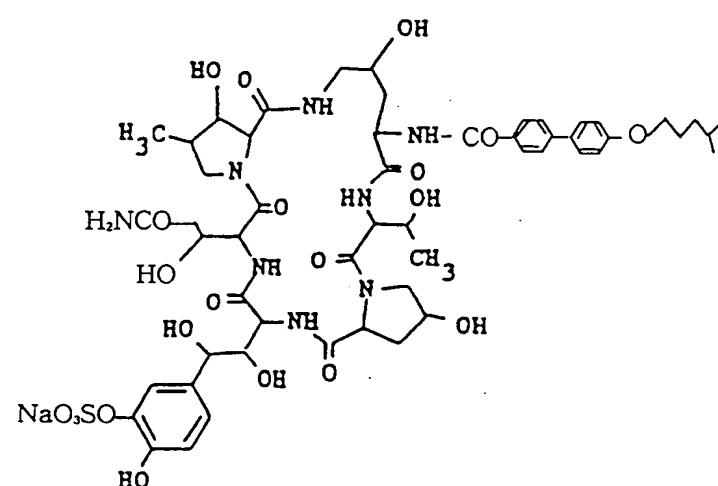
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Example
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Example No.	Formula
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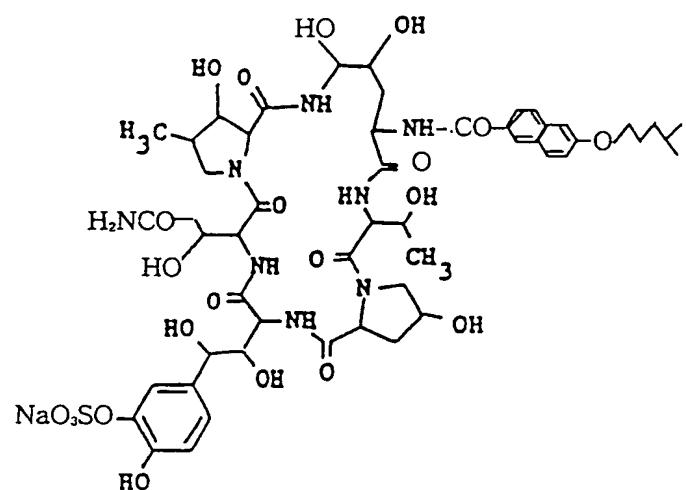
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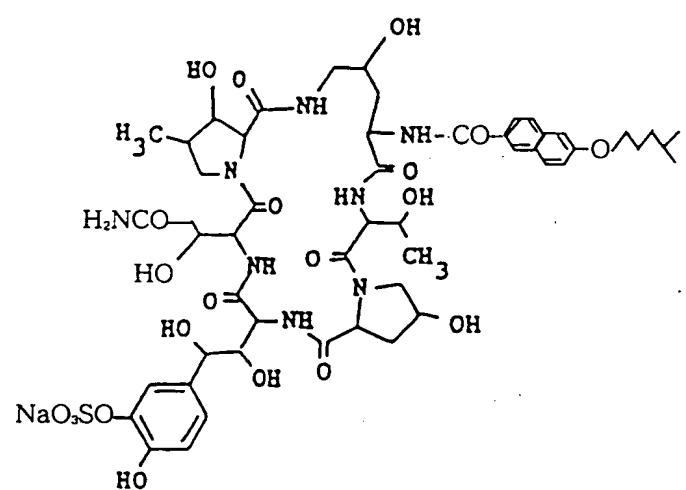
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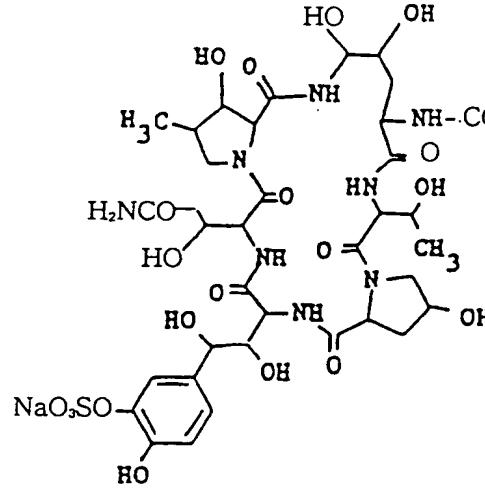
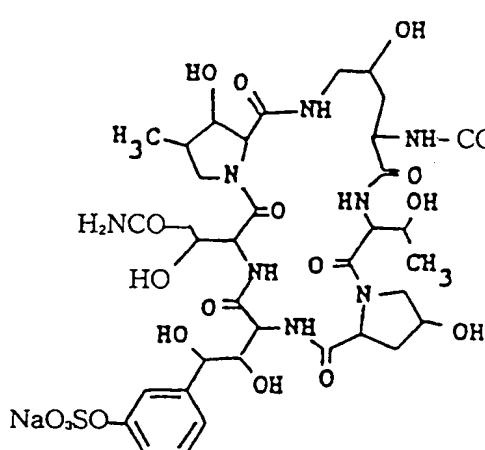
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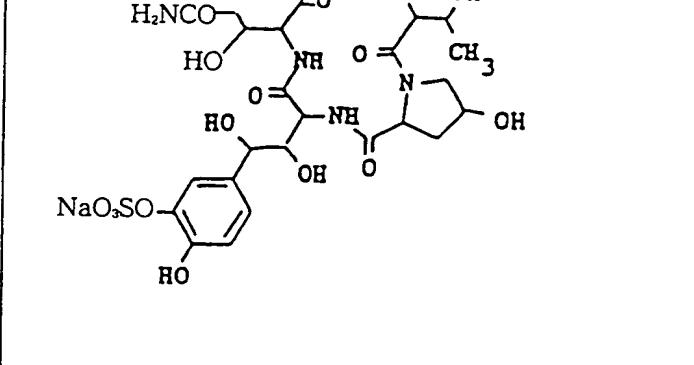
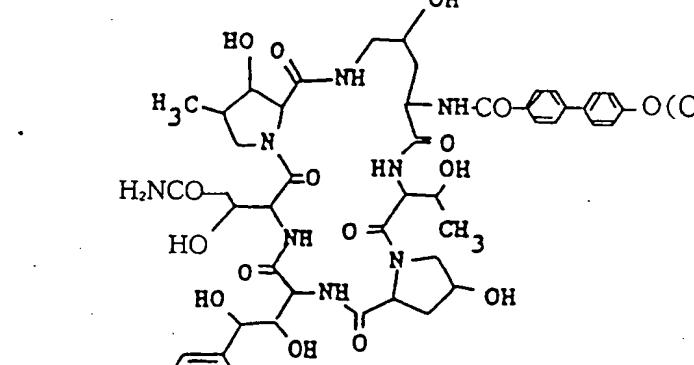


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Example No.	Formula
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Example No.	Formula
	
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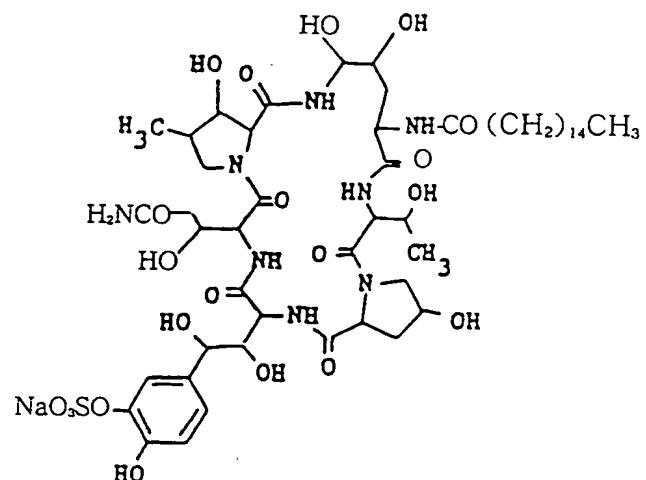
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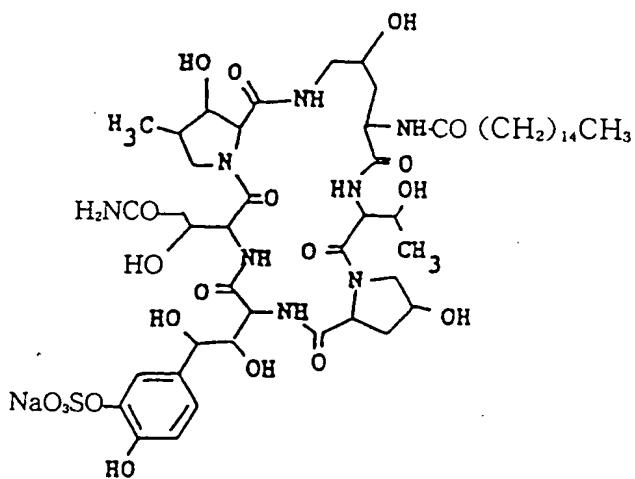
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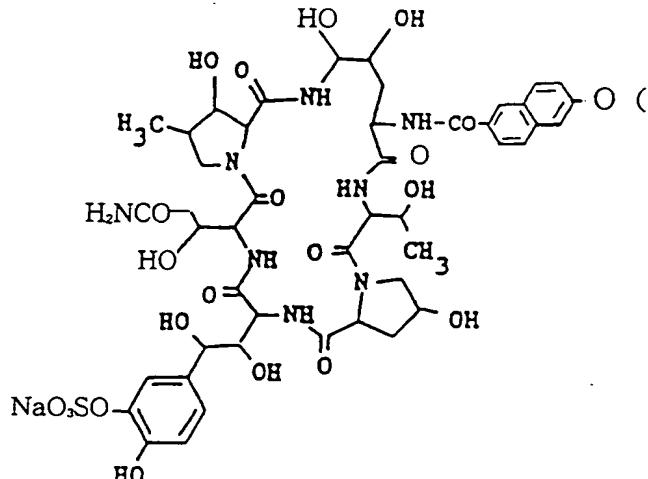
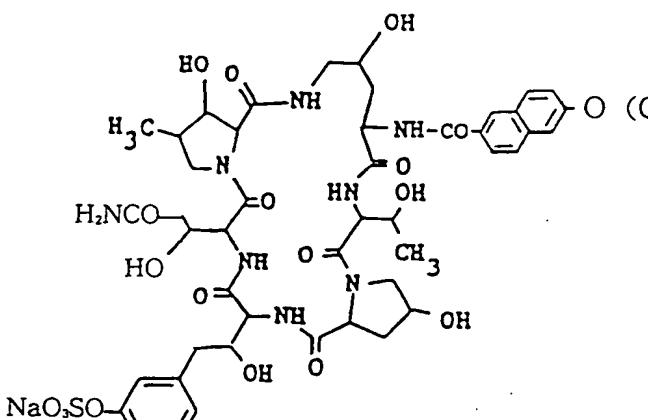
Formula



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Example No.	Formula
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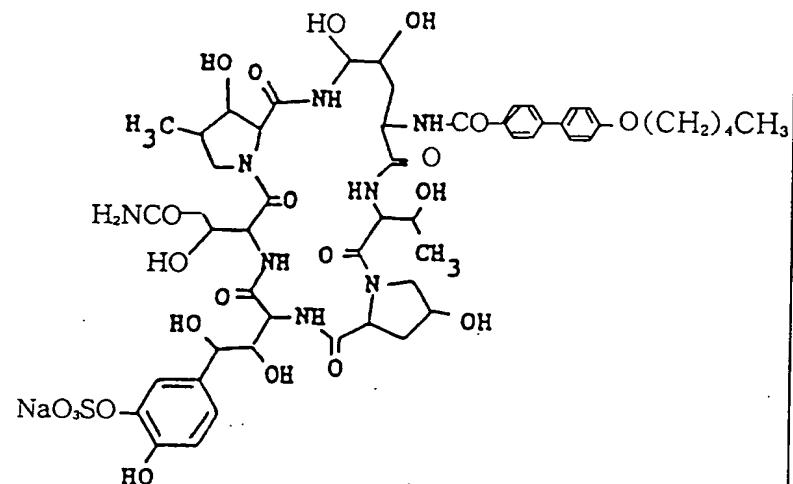
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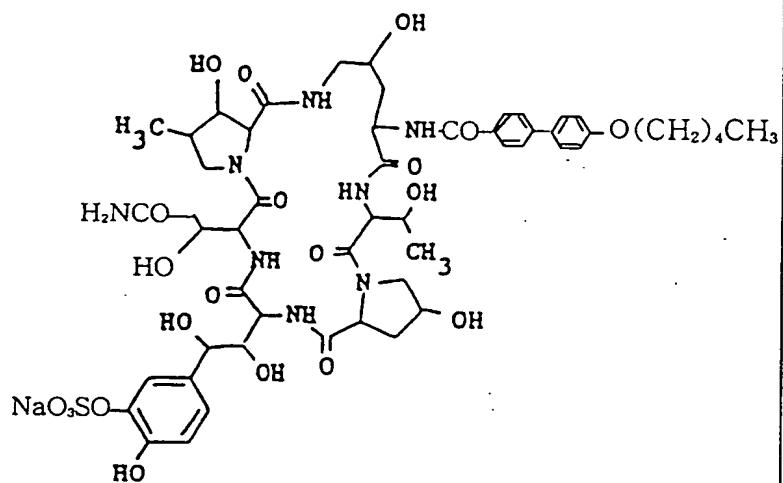
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Example
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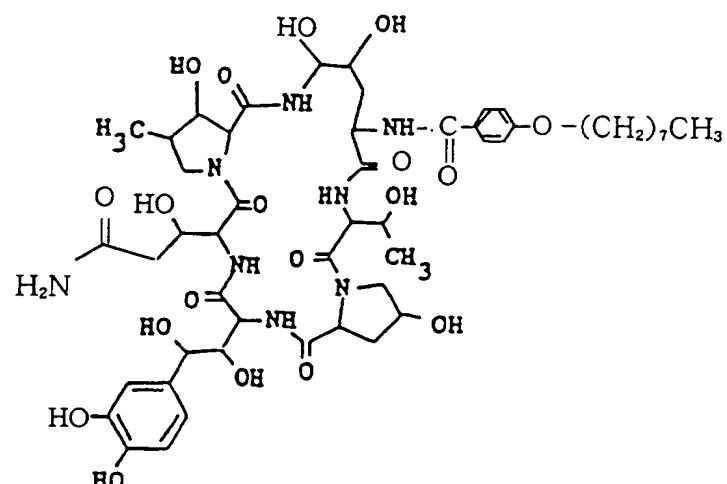
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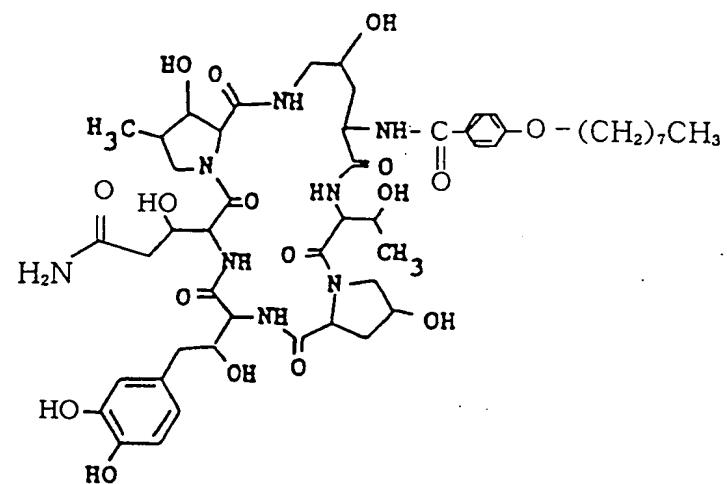
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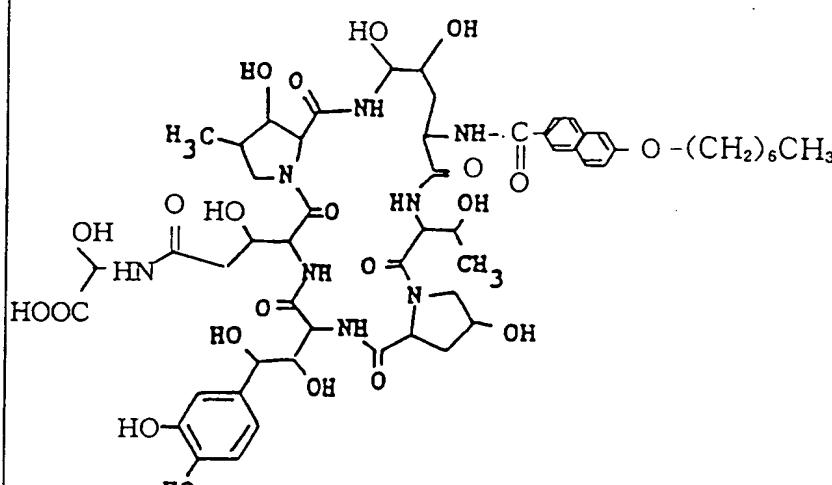
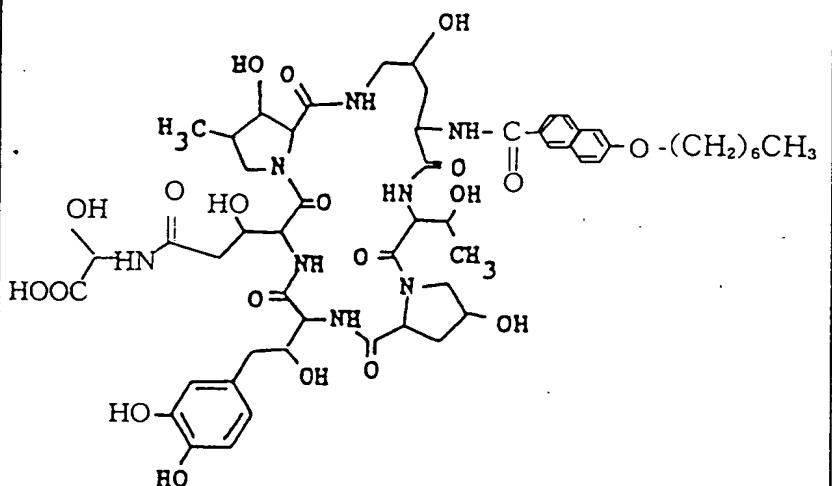
Formula



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Example No.	Formula
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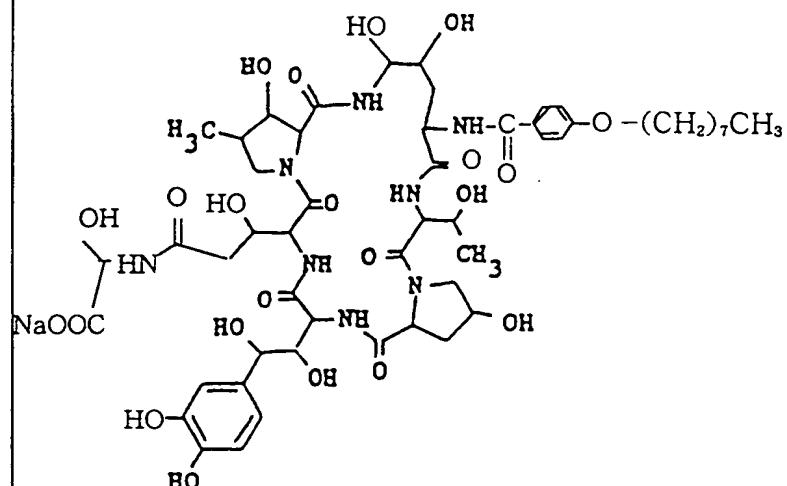
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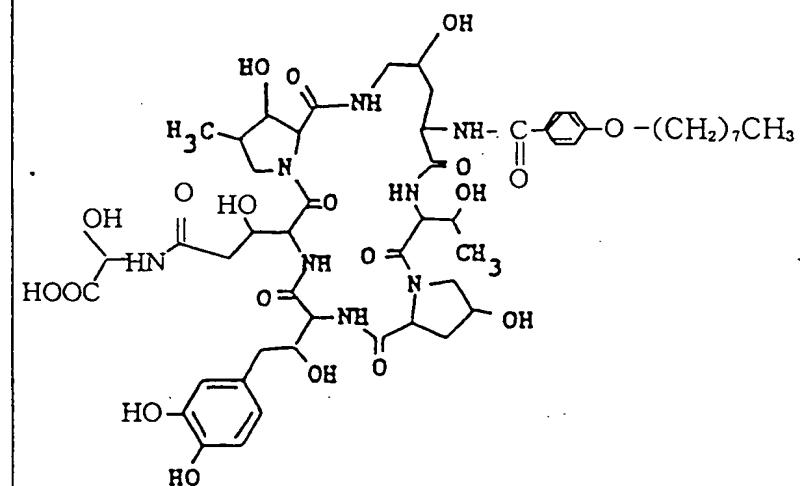
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Example
No.

Formula



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Example No.	Formula
5	<p>Chemical structure of a complex organic molecule, likely a polyamine derivative, featuring a piperazine ring, a piperidine ring, and a cyclopentane ring. It includes multiple hydroxyl groups, amide linkages, and a long-chain sulfonate group (NaO₃SO-phenyl-OH).</p>
16	<p>Chemical structure of a complex organic molecule, identical to the one in Example 5, featuring a piperazine ring, a piperidine ring, and a cyclopentane ring. It includes multiple hydroxyl groups, amide linkages, and a long-chain sulfonate group (NaO₃SO-phenyl-OH).</p>

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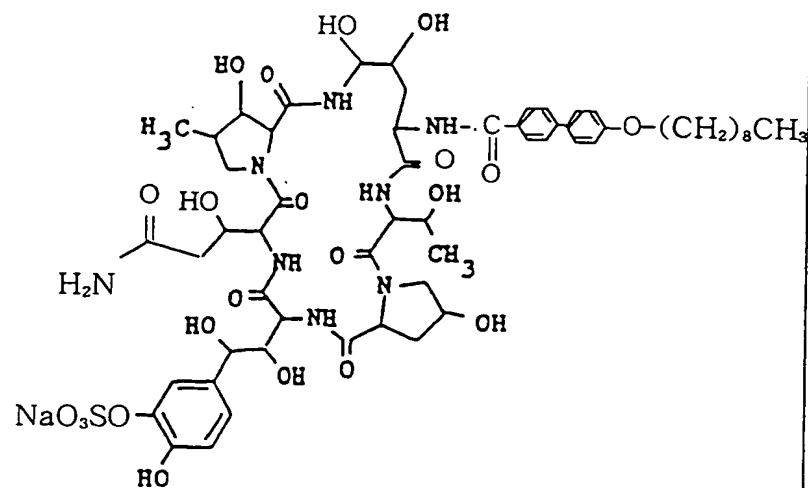
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Example
No.

Formula



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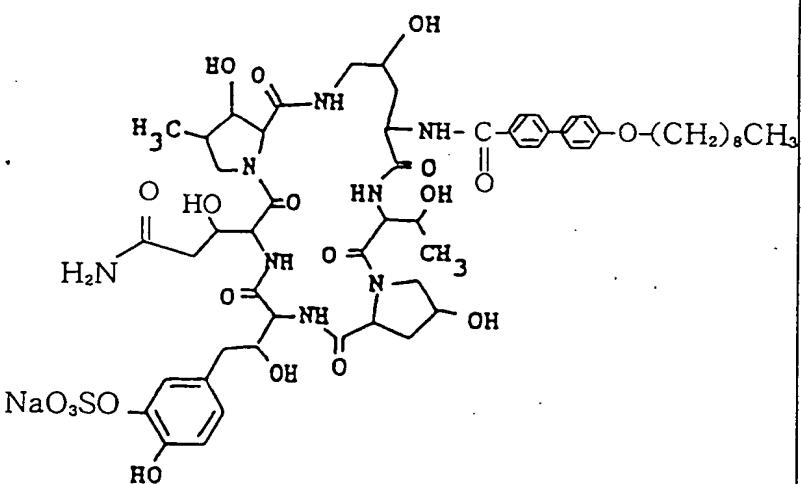
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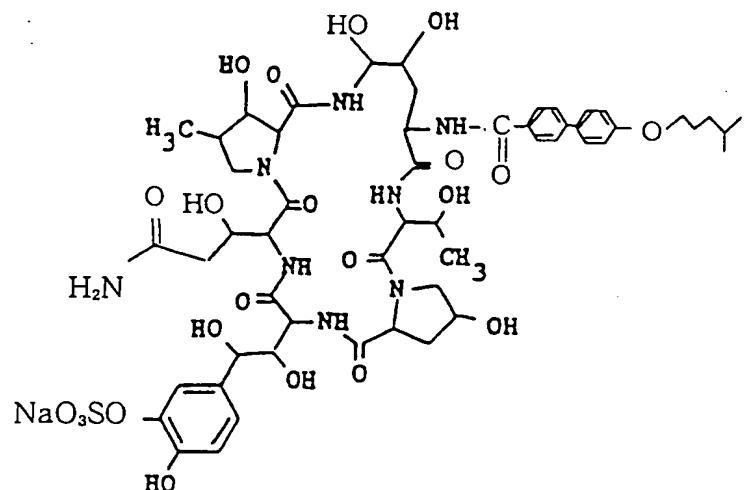
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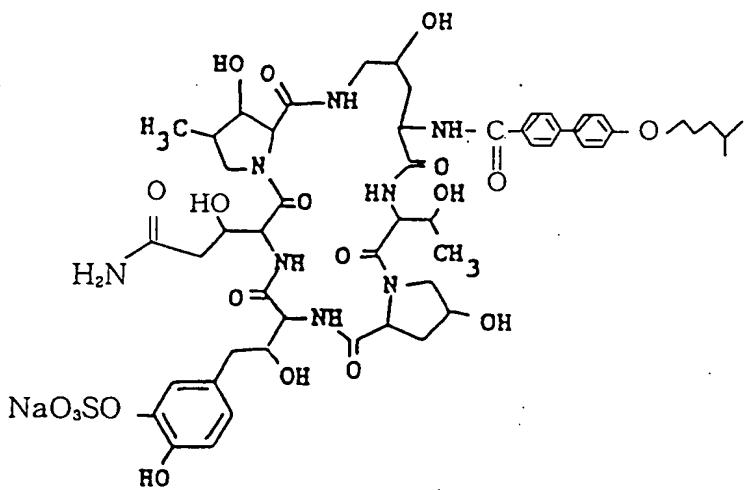
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Example
No.

Formula



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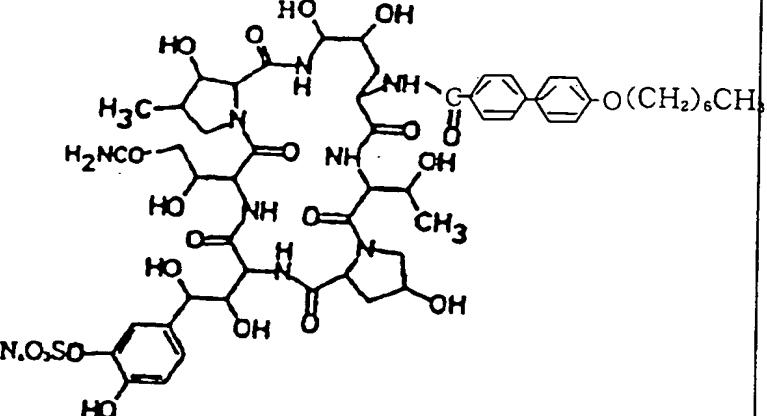
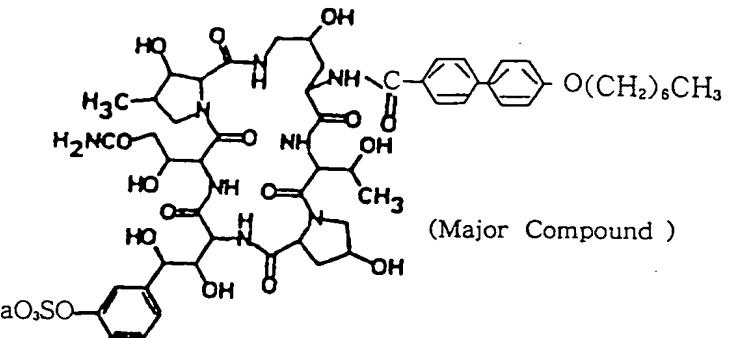
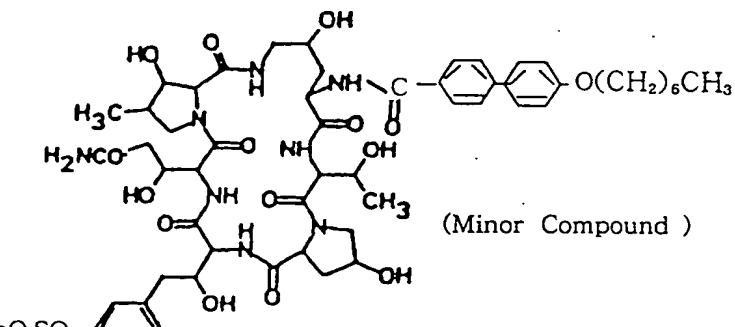
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Example No.	Formula
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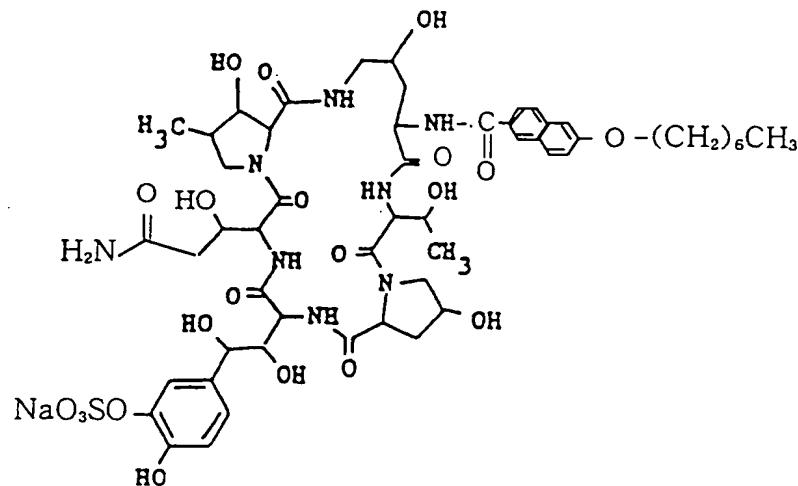
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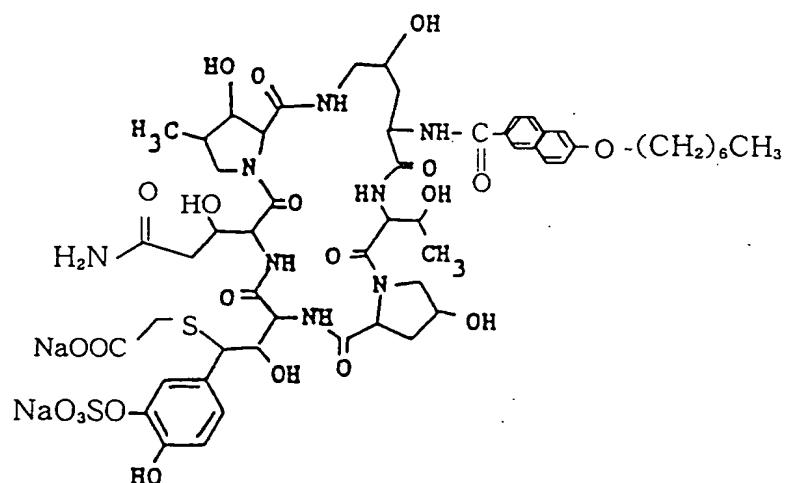
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Example
No.

Formula



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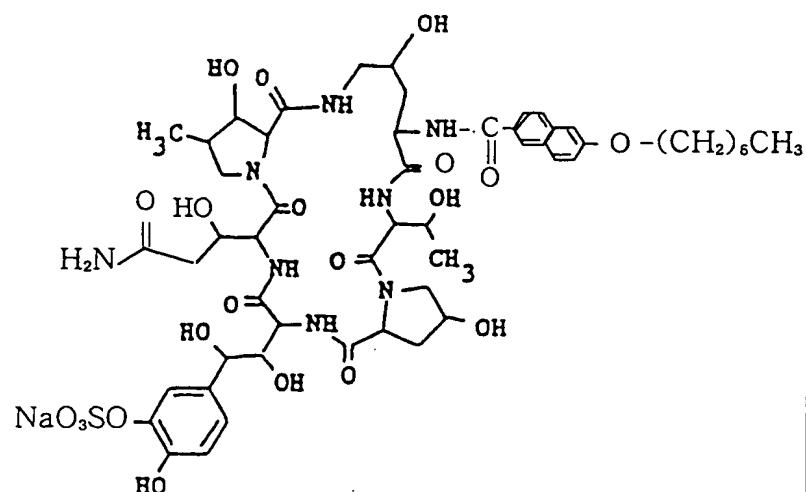
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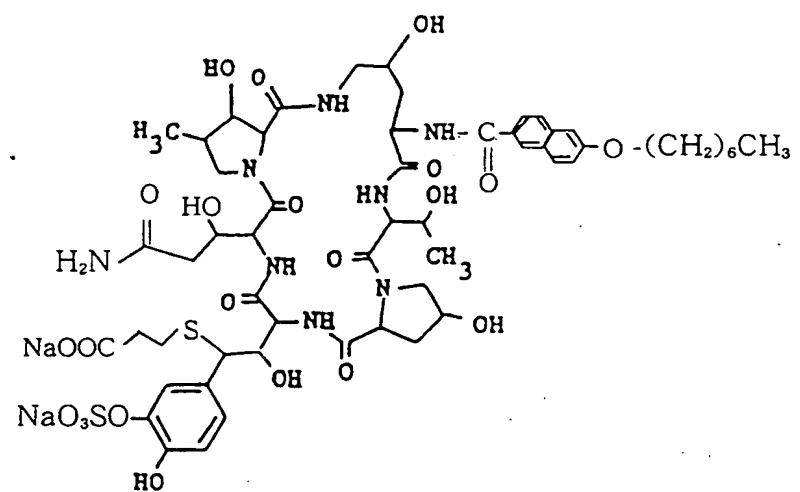
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Example
No.

Formula



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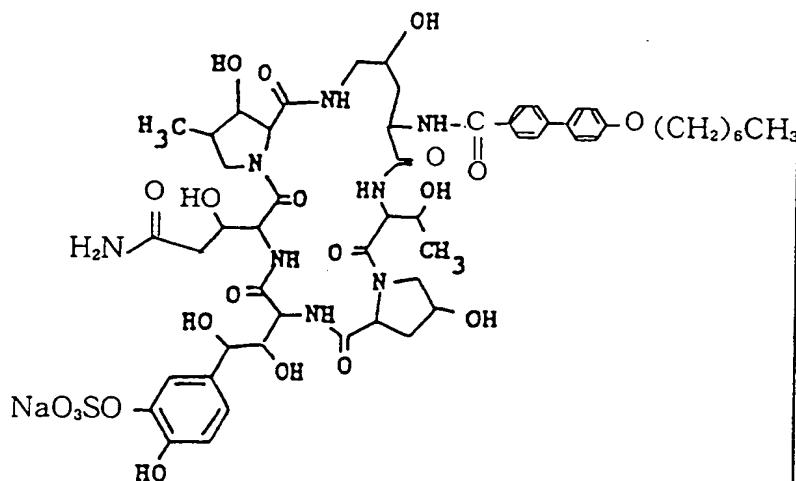
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Example
No.

Formula



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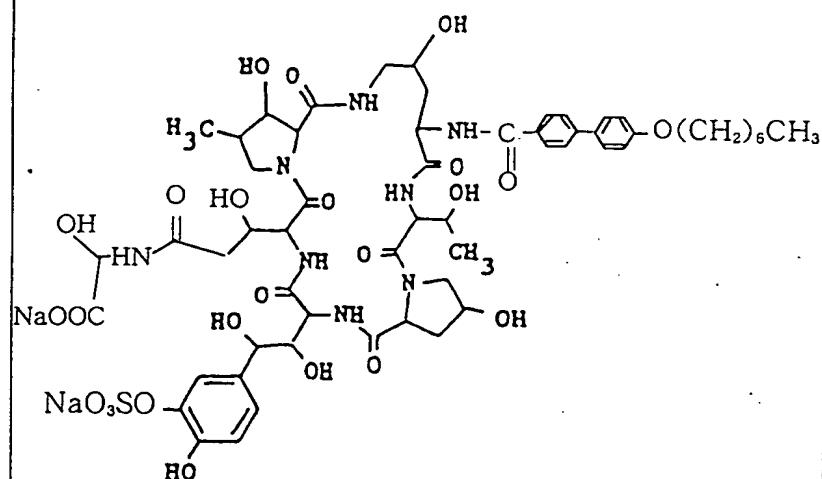
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55 Preparation 10

To a solution of Starting Compound (2.8g) and succinimido 6-heptyloxy-2-naphthoate (1.46g) in N,N-dimethylformamide (28ml) was added 4-(N,N-dimethylamino) pyridine (0.393g) and stirred for 12 hours at

room temperature. The reaction mixture was pulverized with ethyl acetate (140ml). The precipitate was collected by filtration and dried under reduced pressure. The resultant powder was added to water (50ml) and subjected to ion-exchange column chromatography on DOWEX-50WX4 (Trademark : prepared by Dow Chemical) (30ml) and eluted with water. The fractions containing the Object Compound were combined and subjected to column chromatography on ODS YMC-gel (ODS-AM S-50) (Trademark : prepared by Yamamura Chemical Labs) and eluted with 50% aqueous methanol. The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove methanol. The residue was lyophilized to give Object Compound (1.94g).

IR (Nujol) : 3300, 1620cm⁻¹

10 NMR (CD₃OD, δ) : 0.92 (3H, t, J = 6.6Hz), 1.06 (3H, d, J = 6.8Hz), 1.24 (3H, d, J = 6.1Hz), 1.3-1.7 (8H, m), 1.7-2.3 (5H, m), 2.3-2.7 (3H, m), 2.8-2.9 (1H, m), 3.39 (1H, m), 3.7-4.7 (16H, m), 4.99 (1H, d, J = 2Hz), 5.10 (1H, d, J = 3.7Hz), 5.36 (1d, J = 2.9Hz), 6.86 (1H, d, J = 8.3Hz), 7.05 (1H, dd, J = 8.3Hz and 2Hz), 7.17 (1H, dd, J = 8.9Hz and 1.9Hz), 7.23 (1H, d, J = 2Hz), 7.32 (1H, d, J = 1.9Hz), 7.7-7.9 (3H, m), 8.31 (1H, s) FAB-MS : e/z = 1249 (M⁺ + Na)

15

Preparation 11

The Object Compound was obtained according to a similar manner to that of Preparation 10

20 NMR (DMSO-d₆, δ) : 0.91(3H, t, J = 6.6Hz), 0.96(3H, d, J = 7.2Hz), 1.09(3H, d, J = 5.5Hz), 1.25-1.5(4H, m), 1.6-2.6(9H, m), 3.18(1H, m), 3.6-4.6(15H, m), 4.7-5.4(11H, m), 5.52(1H, d, J = 5.8Hz), 6.74(1H, d, J = 8.2Hz), 6.83(1H, d, J = 8.2Hz), 6.86(1H, s), 7.04(2H, d, J = 8.7Hz), 7.06(1H, s), 7.2-7.5(3H, m), 7.68(2H, d, J = 8.7Hz), 7.72(2H, d, J = 8.4Hz), 7.96(2H, d, J = 8.4Hz), 8.12(1H, d, J = 7.9Hz), 8.31 (1H, d, J = 7.1Hz), 8.77(1H, d, J = 7.1Hz), 8.84(1H, s) FAB-MS : e/z = 1247(M⁺ + Na)

25

Analysis : Calcd for C ₅ H ₆ N ₈ NaO ₂₂ S • 6H ₂ O			
	C ; 47.74	H ; 6.12	N ; 8.40
Found	C ; 47.98	H ; 5.92	N ; 8.41

30

Preparation 12

The Object Compound was obtained according to a similar manner to that of Preparation 10

35 NMR (DMSO-d₆, δ) : 0.85(3H, t, J = 6.6Hz), 0.97(3H, d, J = 6.6Hz), 1.08(3H, d, J = 5.5Hz), 1.2-1.55(12H, m), 1.65-2.1(5H, m), 2.1-2.8(4H, m), 3.18(1H, m), 3.65-4.60(15H, m), 4.7-5.2(10H, m), 5.26(1H, d, J = 4.4Hz), 5.53(1H, d, J = 5.8Hz), 6.74(1H, d, J = 8.2Hz), 6.83(1H, d, J = 8.2Hz), 6.86(1H, s), 7.04(2H, d, J = 8.7Hz), 7.06(1H, s), 7.23-7.55(3H, m), 7.68(2H, d, J = 8.7Hz), 7.72(2H, d, J = 8.4Hz), 7.96 (2H, d, J = 8.4Hz), 8.12(1H, d, J = 7.9Hz), 8.31 (1H, d, J = 7.1Hz), 8.77(1H, d, J = 7.1Hz), 8.85(1H, s) FAB-MS : e/z = 1304(M⁺ + Na)

40

Analysis : Calcd for C ₅ H ₇ N ₈ NaO ₂₂ S • 5H ₂ O			
	C ; 49.92	H ; 6.39	N ; 8.17
Found	C ; 49.96	H ; 6.44	N ; 8.23

45

Preparation 13

The Object Compound was obtained according to a similar manner to that of Preparation 10

50 IR(KBr) : 3300, 1668, 1628, 1271, 1216 cm⁻¹
 NMR (DMSO-d₆, δ) : 0.91(6H, d, J = 6.6Hz), 0.96(3H, d, J = 6.7Hz), 1.08(3H, d, J = 5.5Hz), 1.25-1.45(2H, m), 1.5-2.7(10H, m), 3.18(1H, m), 3.72(2H, m), 3.85-4.6(13H, m), 4.73-5.23(10H, m), 5.26(1H, d, J = 4.5Hz), 5.52(1H, d, J = 5.9Hz), 6.74 (1H, d, J = 8.1Hz), 6.83(1H, d, J = 8.1Hz), 6.91(1H, s), 7.05(1H, s), 7.19-7.52(5H, m), 7.84(1H, d, J = 8.7Hz), 7.9-8.0(2H, m), 8.13(1H, d, J = 7.9Hz), 8.33(1H, d, J = 7.1Hz), 8.44(1H, s), 8.80-8.85(1H, s) FAB-MS : e/z = 1235(M⁺ + Na)

Analysis : Calcd for C ₅₂ H ₆₉ N ₈ NaO ₂₁ S • 4H ₂ O			
Found	C ; 48.59 C ; 48.53	H ; 6.04 H ; 6.15	N ; 8.72 N ; 8.54

5

Preparation 14

To a solution of Starting Compound (0.2g) in the mixture of N,N-dimethylformamide (2ml) and acetone (2ml) in presence of molecular sieves 4A was added glyoxylic acid (0.155g) at room temperature. The mixture was stirred at the same temperature for 5 hours. The reaction mixture was pulverized with ethyl acetate (20ml). The precipitate was collected by filtration and dried under reduced pressure to give Object Compound (0.14g).

NMR (CD₃OD, δ) : 0.90 (3H, t, J = 6.6Hz), 1.05 (3H, d, J = 6.7Hz), 1.24 (3H, d, J = 6.0Hz), 1.2-1.6 (10H, m), 1.7-1.9 (2H, m), 1.9-2.2 (3H, m), 2.3-2.6 (3H, m), 2.7-2.9 (1H, m), 3.40 (1H, m), 3.7-4.7 (16H, m), 4.98 (1H, s), 5.09 (1H, brs), 5.31 (1H, brs), 5.40 (1H, s), 6.8-7.2 (4H, m), 7.33 (1H, s), 7.85 (2H, d, J = 8.4Hz).

FAB-MS : e/z = 1287 (M⁺ + Na)

Preparation 15

The Object Compound was obtained according to a similar manner to that of Preparation 14.

IR (Nujol) : 3250, 1610 cm⁻¹

NMR (CD₃OD, δ) : 0.90 (3H, t, J = 6.6Hz), 1.05 (3H, d, J = 6.7Hz), 1.24 (3H, d, J = 6.1Hz), 1.2-1.6 (10H, m), 1.7-1.9 (2H, m), 1.9-2.2 (3H, m), 2.3-2.6 (3H, m), 2.7-2.9 (1H, m), 3.42 (1H, m), 3.7-4.7 (16H, m), 4.98 (1H, s), 5.09 (1H, d, J = 3.8Hz), 5.31 (1H, d, J = 2.8Hz), 5.44 (1H, s), 6.60 (1H, dd, J = 8.1 and 1.9Hz), 6.72 (1H, d, J = 8.1Hz), 6.82 (1H, d, J = 1.9Hz), 6.95 (2H, d, J = 8.8Hz) 7.84 (2H, d, J = 8.8Hz).

FAB-MS : e/z = 1185 (M⁺ + Na)

30

Elemental Analysis : Calcd : for C ₅₂ H ₇₃ N ₈ NaO ₂₂ • 7H ₂ O			
Found :	C ; 47.63 C ; 47.56	H ; 6.76 H ; 6.66	N ; 8.54 N ; 8.35

35

Preparation 16

The Object Compound was obtained according to a similar manner to that of Preparation 14.

NMR (DMSO-d₆, δ) : 0.88 (3H, t, J = 6.6Hz), 0.96 (3H, d, J = 6.7Hz), 1.12 (3H, d, J = 5.9Hz), 1.2-2.0 (13H, m), 2.1-2.7 (4H, m), 3.15 (1H, m), 3.5-4.5 (18H, m), 4.7-5.7 (1H, m), 6.74 (1H, d, J = 8.2Hz), 6.84 (1H, d, J = 8.5Hz), 7.03 (1H, s), 7.22 (1H, d, J = 8.9Hz), 7.36 (1H, s), 7.5 (1H, m), 7.83 (1H, d, J = 8.8Hz), 7.9-8.0 (2H, m), 8.0-8.3 (3H, m), 8.49 (1H, s), 8.65 (1H, m), 8.85 (1H, brs) FAB-MS : e/z = 1324 (M⁺)

45

Elemental Analysis : Calcd : for C ₅₅ H ₇₂ N ₈ Na ₂ O ₂₅ S • 7H ₂ O			
Found :	C ; 45.57 C ; 45.72	H ; 5.98 H ; 5.87	N ; 7.73 N ; 7.69

50

Preparation 17

To a solution of Starting Compound (3.15g) in N,N-dimethylformamide (32ml) was added the mixture of trifluoroacetic acid (0.37ml) and p-toluenesulfonic acid • H₂O (0.5g) at room temperature and stirred for 12hours at the same temperature. The reaction mixture was pulverised with ethyl acetate (300ml). The precipitate was collected by filtration and dried under reduced pressure to give crude product (3.3g). The crude product (0.6g) was added to water (100ml) and purified by preparative HPLC utilizing a C₁₈ μ Bondapak resin (Waters Associates, Inc.) which was eluted with a solvent system comprised of acetonitrile-

pH 3 phosphate buffer (38 : 62) at a flow rate of 80ml/minutes using a Shimadzu LC-8A pump. The column was monitored by a UV detector set at 240nm.

The fractions containing the major compound at retention time of 20.6 minutes were combined and evaporated under reduced pressure to remove acetonitrile. The residue was adjusted to pH 6.5 with

5 saturated sodium bicarbonate aqueous solution and subjected to column chromatography on ODS (YMS-gel ODS-AM S-50) and washed with water and eluted with 80% methanol aqueous solution. The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove methanol.

The residue was lyophilized to give Object Compound (147mg).

10 IR (Nujol) : 3300, 1620cm⁻¹

NMR (DMSO-d₆, δ) : 0.88 (3H, t, J = 6.6Hz), 0.94 (2H, t, J = 6.7Hz), 1.09 (3H, d, J = 5.9Hz), 1.2-2.0 (13H, m), 2.1-2.7 (4H, m), 3.20 (1H, m), 3.6-4.5 (18H, m), 4.7-5.2 (10H, m), 5.4-5.6 (2H, m), 6.44 (1H, d, J = 8.2Hz), 6.6 (1H, d, J = 8.2Hz), 6.70 (1H, s), 7.1-7.5 (3H, m), 7.7-8.2 (6H, m), 8.3-8.9 (5H, m)

15 Example 1

To a solution of Starting Compound (1g) in trifluoroacetic acid (5ml) in presence of molecular sieves 4 Å was added sodium cyanoborohydride (0.285g) at ambient temperature. The mixture was stirred for an hour at the same temperature. The reaction mixture was added to water (10ml) under ice-cooling and

20 adjusted to pH 7 with 1N aqueous sodium hydroxide. The solution was subjected to ion-exchange column chromatography on DOWEX-50WX4 (Na⁺ Type) (30ml) and eluted with water. The fractions containing the Object Compound were combined and purified by preparative HPLC utilizing a C18 μ Bondapak resin (Waters Associates Inc.) which was eluted with a solvent system composed of acetonitrile -pH3 phosphate buffer (39 : 61) at a flow rate of 80ml/minute using a Shimadzu LC-8A pump. The column was monitored by

25 a UV detector set at 240nm. The fractions containing the first eluted compound at retention time of 17.1 minute were combined and evaporated under reduced pressure to remove acetonitrile. The residue was subjected to column chromatography on ODS (YMC-gel ODS -AMS-50) and washed with water and eluted with 80% aqueous methanol. The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove methanol.

30 The residue was lyophilized to give Object Compound 1 (318mg).

IR (Nujol) : 3300, 1610, 1230cm⁻¹

NMR (DMSO-d₆, δ) : 0.86 (3H, t, J = 6.6Hz), 0.95 (3H, d, J = 6.7Hz), 1.09 (3H, d, J = 5.9Hz), 1.2-2.0 (15H, m), 2.1-2.6 (4H, m), 3.00 (1H, m), 3.19 (1H, m), 3.74 (2H, m), 3.8-4.5 (14H, m), 4.6-5.4 (10H, m), 6.73 (1H, d, J = 8.2Hz), 6.80 (1H, s), 6.84 (1H, dd, J = 8.2 and 1.9Hz), 6.97 (2H, d, J = 8.8Hz), 7.06 (1H, d, J = 1.9Hz), 7.25 (1H, s), 7.44 (2H, m), 7.69 (1H, m), 7.83 (2H, d, J = 8.8Hz), 8.09 (1H, d, J = 7Hz), 8.44 (1H, d, J = 7Hz), 8.85 (1H, s)

FAB-MS : e/z = 1197 (M⁺ + Na)

40

Elemental Analysis : Calcd : for C ₅₀ H ₇₁ N ₈ NaO ₂₁ S • 6H ₂ O			
Found :	C ; 46.79 C ; 46.84	H ; 6.51 H ; 6.27	N ; 8.73 N ; 8.73

45 The second eluted compound at retention time of 20.2 min. was obtained according to a similar manner to that of the first eluted compound, Object Compound 2 (263mg).

IR (Nujol) : 3250, 1600, 1490, 1240cm⁻¹

NMR (DMSO-d₆, δ) : 0.86 (3H, t, J = 6.6Hz), 0.95 (3H, d, J = 6.7Hz), 1.09 (3H, d, J = 5.9Hz), 1.2-2.0 (15H, m), 2.1-2.6 (6H, m), 2.96 (1H, m), 3.21 (1H, m), 3.6-4.5 (16H, m), 4.6-5.4 (8H, m), 6.73 (1H, d, J = 8.2Hz), 6.77 (1H, dd, J = 8.2 and 1.9Hz), 6.81 (1H, s), 6.97 (2H, d, J = 8.8Hz), 6.98 (1H, d, J = 1.9Hz), 7.25 (1H, s), 7.39 (2H, m), 7.39 (1H, m), 7.74 (1H, m), 7.84 (2H, d, J = 8.8Hz), 8.14 (1H, d, J = 7Hz), 8.46 (1H, m), 8.72 (1H, s)

FAB-MS : e/z = 1181 (M⁺ + Na)

Elemental Analysis : Calcd : for C ₅₀ H ₇ N ₈ NaO ₂₀ S • 6H ₂ O			
Found :	C ; 47.38 C ; 47.46	H ; 6.60 H ; 6.62	N ; 8.84 N ; 8.83

5

Example 2

To suspension of Starting Compound (10.4g) and sodium cyanoborohydride (1.67g) in dichloromethane (52ml) was gradually added tetrahydrofuran (104ml) at 5 °C. The mixture was stirred at 40 °C for 1 hour. The reaction mixture was evaporated under reduced pressure. To the residue was added water, adjusted to pH 8.5 with 1N sodium hydroxide and subjected to column chromatography on ODS (YMC-gel, ODS-AM S-50) and eluted with 60% acetonitrile aq. The fractions containing the crude product were combined and evaporated under reduced pressure to remove acetonitrile. The residue was lyophilized to give crude powder (10.1g). The crude powder was purified by column chromatography on silica gel using dichloromethane /acetic acid/methanol/water (4:1:1:1) as eluent. The fractions containing the Object Compound were combined and evaporated under reduced pressure. To the residue was added water, adjusted to pH 8.5 with 1N sodium hydroxide and subjected to column chromatography on ODS (YMC-gel ODS-AM S-50) and eluted with 60% acetonitrile aq. The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove acetonitrile. The residue was lyophilized to give Object Compound (2.25g).

IR (Nujol) : 3250, 1600, 1490, 1240 cm⁻¹

NMR (DMSO-d₆, δ) : 0.86(3H, t, J = 6.6Hz), 0.95(3H, d, J = 6.7Hz), 1.09(3H, d, J = 5.9Hz), 1.2-2.0(15H, m), 2.1-2.6(6H, m), 2.96(1H, m), 3.21(1H, m), 3.6-4.5(16H, m), 4.6-5.4(8H, m), 6.73(1H, d, J = 8.2Hz), 6.77(1H, dd, J = 8.2 and 1.9Hz), 6.81 (1H, s), 6.97(2H, d, J = 8.8Hz), 8.14 (1H, d, J = 7Hz), 8.46 (1H, m), 8.72 (1H, s)

FAB-MS : e/z = 1181 (M⁺ + Na)

Elemental Analysis : Calcd : for C ₅₀ H ₇ N ₈ NaO ₂₀ S • 6H ₂ O			
Found :	C;47.38 C;47.46	H;6.60 H;6.62	N;8.84 N;8.83

30

Example 3

To a suspension of Starting Compound (1g) and sodium cyanoborohydride (0.256g) in dichloromethane (10ml) was gradually added tetrahydrofuran (5ml) at 5 °C. The mixture was stirred at 5 °C for 1.5 hours. The reaction mixture was pulverized with diisopropyl ether (200ml). The precipitate was collected by filtration and dried under reduced pressure. The powder was purified by column chromatography on silica gel using dichloromethane /acetic acid/methanol/water (3:1:1:1) as eluent. The fractions containing the Object Compound were combined and evaporated under reduced pressure. To the residue was added water, adjusted to pH 8.5 with 1N sodium hydroxide and subjected to column chromatography on ODS (YMC-gel ODS-AM S-50) and eluted with 60% acetonitrile aq. The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove acetonitrile. The residue was lyophilized to give Object Compound (0.21g).

IR (Nujol) : 3300, 1620, 1270 cm⁻¹

NMR (DMSO-d₆, δ) : 0.88(3H, t, J = 6.6Hz), 0.96(3H, d, J = 6.8Hz), 1.12(3H, d, J = 5.9Hz), 1.2-1.55(8H, m), 1.65-2.6(9H, m), 2.97(1H, m), 3.20(1H, m), 3.26-3.52 (1H, m), 3.73(2H, m), 3.86-4.63(14H, m), 4.63-5.36(9H, m), 6.74(1H, d, J = 8Hz), 6.83(1H, dd, J = 1.5 and 8Hz), 6.92(1H, s), 7.06(1H, d, J = 1.5Hz), 7.2-7.35(2H, m), 7.35-7.6(3H, m), 7.68(1H, m), 7.8-8.0(3H, m), 8.09(1H, d, J = 8Hz), 8.42(1H, s), 8.67(1H, d, J = 6.9Hz), 8.71(1H, s)

FAB-MS : e/z = 1233 (M⁺ + Na)

55

Elemental Analysis: Calcd : for C ₅₃ H ₇₁ N ₈ NaO ₂₁ S•6H ₂ O			
Found :	C;48.25 C;48.06	H;6.34 H;6.29	N;8.49 N;8.31

5

Example 4

10 The Object Compound was obtained according to a similar manner to that of Example 3.
 NMR (DMSO-d₆, δ) : 0.9-1.1(6H, m), 1.11(3H, d, J = 5.8Hz), 1.3-1.5(4H, m), 1. 7-2.1(5H, m), 2.2-2.4(4H, m), 2.9-3.1(2H, m), 3.4(1H, m), 3.7-4.5(16H, m), 4. 7-5.3(9H, m), 6.7-7.1(6H, m), 7.22(1H, s), 7.41(1H, d, J = 8.5Hz), 7.6-7.8(5H, m), 7.93(2H, d, J = 8.2Hz), 8.08(1H, d, J = 8.5Hz), 8.60(1H, d, J = 7.1Hz), 8.85(1H, s)
 FAB-MS : e/z = 1231(M⁺ + Na)

15

Elemental Analysis : Calcd for C ₅₃ H ₆₉ N ₈ NaO ₂₁ S•3.5H ₂ O			
Found	C ; 50.04 C ; 50.05	H ; 6.02 H ; 6.25	N ; 8.87 N ; 8.81

20

Example 5

25 The Object Compound was obtained according to a similar manner to that of Example 3.
 IR(Nujol) : 3300, 1635, 1247, 1047 cm⁻¹
 NMR (DMSO-d₆, δ) : 0.86(3H, t, J = 6.7Hz), 0.95(3H, d, J = 6.7Hz), 1.10(3H, d, J = 5.8Hz), 1.2-1.6(12H, m), 1.6-2.6(9H, m), 3.0(1H, m), 3.2(1H, m), 3.4(1H, m), 3.74(2H, m), 3.83-4.6(14H, m), 4.65-5.4(9H, m), 6.73(1H, d, J = 8.2Hz), 6.81(1H, dd, J = 1.7 and 8.2Hz), 6.89(1H, s), 7.03(2H, d, J = 8.8Hz), 7.05(1H, d, J = 1.7Hz), 7.23-(1H, s), 7.41(2H, m), 7.6-7.8(1H, m), 7.67(2H, d, J = 8.8Hz), 7.71(2H, d, J = 8.5Hz), 7.93(2H, d, J = 8.5Hz), 8.05(1H, d, J = 8Hz), 8.62(1H, d, J = 6.7Hz), 8.84 (1H, s)
 FAB-MS : e/z = 1287(M⁺ + Na)

35

Elemental Analysis : Calcd for C ₅₇ H ₇₇ N ₈ NaO ₂₁ S•8H ₂ O			
Found	C ; 48.57 C ; 48.41	H ; 6.65 H ; 6.26	N ; 7.95 N ; 7.89

Example 6

40 The Object Compound was obtained according to a similar manner to that of Example 3.
 IR(Nujol) : 3300, 1633, 1247, 1047 cm⁻¹
 NMR (DMSO-d₆, δ) : 0.90(6H, d, J = 6.5Hz), 0.95(3H, d, J = 6.8Hz), 1.10(3H, d, J = 5.7Hz), 1.31(2H, q, J = 7.5Hz), 1.45-2.6(12H, m), 2.97(1H, m), 3.18(1H, m), 3.40 (1H, m), 3.74(2H, m), 3.83-5.4(23H, m), 6.72(1H, d, J = 8.2Hz), 6.81(1H, d, J = 8.2Hz), 6.89(1H, s), 7.03(2H, d, J = 8.9Hz), 7.06(1H, s), 7.2-7.6(3H, m), 7.6-7.8 (1H, m), 7.67(2H, d, J = 8.9Hz), 7.71(2H, d, J = 8.4Hz), 7.93(2H, d, J = 8.4Hz), 8.05 (1H, m), 8.61(1H, d, J = 6.7Hz), 8.84(1H, s)
 FAB-MS : e/z = 1245(M⁺ + Na)

50

Elemental Analysis : Calcd for C ₅₄ H ₇₁ N ₈ NaO ₂₀ S•7H ₂ O			
Found	C ; 48.64 C ; 48.72	H ; 6.43 H ; 6.25	N ; 8.40 N ; 8.26

55

Example 7

The Object Compound was obtained according to a similar manner to that of Example 3.

IR(Nujol) : 3300, 1625 cm^{-1}

5 NMR (DMSO-d₆, δ) : 0.91(6H, d, $J = 6.6\text{Hz}$), 0.95(3H, d, $J = 6.7\text{Hz}$), 1.10(3H, d, $J = 5.4\text{Hz}$), 1.28-1.47(2H, m), 1.5-2.6(10H, m), 2.97(1H, m), 3.18(1H, m), 3.40(1H, m), 3.74 (2H, m), 3.87-5.45 (23H, m), 6.71(1H, d, $J = 8.3\text{Hz}$), 6.79(1H, d, $J = 8.3\text{Hz}$), 6.91(1H, s), 7.05(1H, s), 7.18-7.58(5H, m), 7.69(1H, m), 7.8-8.0(3H, m), 8.42 (1H, s), 8.60(1H, d, $J = 6.7\text{Hz}$)

FAB-MS : $e/z = 1219(\text{M}^+ + \text{Na})$

10

Elemental Analysis : Calcd for C ₅₂ H ₆₉ N ₈ NaO ₂₁ S • 7H ₂ O			
Found	C ; 47.20 C ; 47.43	H ; 6.32 H ; 6.37	N ; 8.47 N ; 8.34

15

Example 8

20 The Object Compound was obtained according to a similar manner to that of Example 3.

IR(Nujol) : 3300, 1620, 1272 cm^{-1}

NMR (DMSO-d₆, δ) : 0.89(3H, t, $J = 6.7\text{Hz}$), 0.95(3H, d, $J = 6.9\text{Hz}$), 1.11(3H, d, $J = 5.8\text{Hz}$), 1.25-1.6(6H, m), 1.65-2.6(9H, m), 2.97(1H, m), 3.20(1H, m), 3.40(1H, m), 3.74(2H, m), 3.87-5.53(23H, m), 6.73(1H, d, $J = 8.2\text{Hz}$), 6.82(1H, dd, $J = 1.7$ and 8.2Hz), 6.91(1H, s), 7.04(1H, d, $J = 1.7\text{Hz}$), 7.23(1H, dd, $J = 2$, 8.9Hz), 7.29 (1H, s), 7.37(1H, d, $J = 2\text{Hz}$), 7.47(2H, m), 7.70(1H, m), 7.8-8.0(3H, m), 8.07(1H, d, $J = 8\text{Hz}$), 8.42(1H, s), 8.64(1H, d, $J = 6.9\text{Hz}$), 8.85(1H, s)

FAB-MS : $e/z = 1219(\text{M}^+ + \text{Na})$

30

Analysis : Calcd for C ₅₂ H ₆₉ N ₈ NaO ₂₁ S • 6H ₂ O			
Found	C ; 47.85 C ; 48.03	H ; 6.25 H ; 6.01	N ; 8.58 N ; 8.39

35 Example 9

The Object Compound was obtained according to a similar manner to that of Example 3.

IR(Nujol) : 3300, 1620, 1247, 1045 cm^{-1}

NMR (DMSO-d₆, δ) : 0.89(3H, t, $J = 6.9\text{Hz}$), 0.96(3H, d, $J = 6.8\text{Hz}$), 1.11(3H, d, $J = 5.9\text{Hz}$), 1.2-1.55(6H, m), 1.6-2.1(5H, m), 2.1-2.6(4H, m), 2.99(1H, m), 3.19 (1H, m), 3.40(1H, m), 3.74(2H, m), 3.85-4.6(14H, m), 4.65-5.40(9H, m), 6.74 (1H, d, $J = 8.2\text{Hz}$), 6.83(1H, dd, $J = 1.7$ and 8.2Hz), 6.89(1H, s), 7.03(2H, d, $J = 8$, 8Hz), 7.06(1H, d, $J = 1.7\text{Hz}$), 7.23(1H, s), 7.43(2H, m), 7.6-7.8(1H, m), 7.67(2H, d, $J = 8.8\text{Hz}$), 7.71(2H, d, $J = 8.4\text{Hz}$), 7.93(2H, d, $J = 8.4\text{Hz}$), 8.06(1H, d, $J = 8.0\text{Hz}$), 8.58(1H, d, $J = 6.7\text{Hz}$), 8.84(1H, s)

FAB-MS : $e/z = 1245(\text{M}^+ + \text{Na})$

45

Elemental Analysis : Calcd for C ₅₄ H ₇₁ N ₈ NaO ₂₁ S • 5H ₂ O			
Found	C ; 49.39 C ; 49.51	H ; 6.22 H ; 6.22	N ; 8.53 N ; 8.53

50

Example 10

55 The Object Compound was obtained according to a similar manner to that of Example 3.

IR(KBr) : 3300, 1632, 1249, 1047 cm^{-1}

NMR (DMSO-d₆, δ) : 0.85(3H, t, $J = 6.7\text{Hz}$), 0.94(3H, d, $J = 6.7\text{Hz}$), 1.03(3H, d, $J = 5.7\text{Hz}$), 1.1-2.6(35H, m), 2.94(1H, m), 3.17(1H, m), 3.4(1H, m), 3.72(2H, m), 3.9-4.5(12H, m), 4.6-5.35(9H, m), 6.72(1H, d, $J = 8.2\text{Hz}$), 6.79(1H, s), 6.81(1H, d, $J = 8.2\text{Hz}$), 7.04(1H, s), 7.19(1H, s), 7.34(2H, m), 7.67(1H, m), 8.05(2H, m), 8.83(1H,

s)

FAB-MS : e/z = 1203(M⁺ + Na)

5

Elemental Analysis : Calcd for C ₅ H ₈ N ₈ NaO ₂₀ S•4H ₂ O			
Found	C ; 48.87 C ; 49.17	H ; 7.16 H ; 7.08	N ; 8.94 N ; 8.81

10

Example 11

To a solution of Starting Compound(3g) in tetrahydrofuran(30ml) was gradually added sodium cyanoborohydride (0.461g) at 5 °C. The mixture was stirred at 40 °C for 1hour. The reaction mixture was pulverized with diisopropyl ether (150ml). The precipitate was collected by filtration and dried under reduced pressure. The powder was purified by column chromatography on silica gel using dichloromethane /acetic acid/ methanol /water(3:1:1:1) as eluent. The fractions containing the Object Compound were combined and evaporated under reduced pressure. To the residue was added water, adjusted to pH 8.5 with 1N sodium hydroxide and subjected to column chromatography on ODS(YMC-gel ODS-AM S-50) and eluted with 60% acetonitrile aq. The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove acetonitrile. The residue was lyophilized to give Object Compound (0.28g).

NMR (DMSO-d₆, δ) : 0.88(3H, t, J = 6.8Hz), 0.97(3H, d, J = 6.7Hz), 1.10(3H, d, J = 5.9Hz), 1.2-1.55(8H, m), 1.7-2.6(10H, m), 2.98(1H, m), 3.22(1H, m), 3.46(1H, m), 3.64-4.62(15H, m), 4.7-5.5(9H, m), 6.71(1H, d, J = 8.2Hz), 6.77(1H, dd, J = 1.7 and 8.2Hz), 6.91(1H, s), 6.98(1H, d, J = 1.7Hz), 7.2-7.35(2H, m), 7.35-7.52(2H, m), 7.6 (1H, m), 7.74(1H, m), 7.82-8.05(3H, m), 8.18(1H, d, J = 8Hz), 8.43(1H, s), 8.67(1H, d, J = 6.9Hz), 8.72(1H, s)

FAD-MS : e/z 1217 (M⁺ + Na)

30

Elemental Analysis : Calcd : for C ₅ H ₇ N ₈ NaSO ₂₀ •5H ₂ O			
Found :	C ; 49.53 C ; 49.66	H ; 6.35 H ; 6.51	N ; 8.72 N ; 9.10

35

Example 12The Object Compound was obtained according to a similar manner to that of Example 11.

NMR (DHSO-d₆, δ) : 0.9-1.2(9H, m), 1.3-1.5(4H, m), 1.7-2.4(10H, m), 2.9-3.4(4H, m), 3.5-3.6(1H, m), 3.7-4.1(7H, m), 4.1-4.5(8H, m), 4.7-4.8(1H, m), 4.8-4.9(2H, d, J = 6.4Hz), 5.0-5.1(1H, m), 5.1-5.3(3H, m), 5.3-5.4(1H, m), 6.7-7.1 (6H, m), 7.23(1H, s), 7.45(1H, d, J = 7.8Hz), 7.5-7.8(5H, m), 7.94(2H, d, J = 8.6Hz), 8.17(1H, d, J = 7.8Hz), 8.61(1H, d, J = 6.4Hz), 8.71(1H, s) FAD-MS : e/z = 1215(M⁺ + Na)

Example 13

45

To a solution of Starting Compound (5.31g) in trifluoroacetic acid (100ml) was gradually added sodium cyanoborohydride (0.92g) at ambient temperature. The mixture was stirred for an hour at the same temperature.

The reaction mixture was pulverized with diisopropyl ether (750ml). The precipitate was collected by filtration and dried under reduced pressure. The powder was subjected to column chromatography on silica gel (11) and eluted with dichloromethane /acetic acid/ methanol/water (3:1:1: 1). The fractions containing the Object Compound were combined and evaporated under reduced pressure to give Object Compound (2.36g).

IR(KBr) : 3300, 1628, 1201cm⁻¹

NMR(DMSO-d₆, δ) : 0.86(3H, t, J = 6.6Hz), 0.95(3H, d, J = 6.7Hz), 1.08(3H, d, J = 5.9Hz), 1.2-1.5(10H, m), 1.58-2.6(10H, m), 2.96(1H, m), 3.0-3.6(2H, m), 3.65-4.53(15H, m), 4.6-5.46 (9H, m), 6.40(1H, d, J = 8.1Hz), 6.59(1H, s), 6.61(1H, d, J = 8.1Hz), 6.87(1H, s), 6.96(2H, d, J = 8.9Hz), 7.15-7.75(3H, m) 7.84(2H, d, J = 8.9Hz), 8.12(1H, m), 8.43(1H, m)

FAD-MS : e/z = 1079(M⁺ + Na)Example 14

5 To a solution of Starting Compound (1.469g) in trifluoroacetic acid (14.7ml) was gradually added sodium cyanoborohydride (0.452g) at ambient temperature. The mixture was stirred for an hour at the same temperature. The reaction mixture was pulverized with diethyl ether (400ml). The precipitate was collected by filtration and dried under reduced pressure. The powder was added to water (50ml) and adjusted to pH7 with saturated NaHCO₃ aqueous solution, and subjected to column chromatography on ODS (YMC-gel 10 ODS-AM S-50) and eluted with 60% aqueous methanol. The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove methanol. The residue was lyophilized to give Object Compound (0.85g).

IR(Nujol) : 3250, 1600, 1060cm⁻¹

15 NMR(DMSO-d₆, δ) : 0.86(3H, t, J = 6.7Hz), 0.96(3H, d, J = 6.6Hz), 1.16(3H, d, J = 5.7Hz), 1.23-1.63(8H, m), 1.63-2.1(5H, m), 2.1-2.67(5H, m), 2.95-3.67(3H, m), 3.67-4.78(16H, m), 4.78-6.11(10H, m), 6.38(1H, d, J = 7.9Hz), 6.5-6.7(2H, m), 7.1-7.7(4H, m), 7.7-8.32(5H, m), 8.32-9.1(5H, m)

FAB-MS:e/z = 1115.6(M⁺-HCOCOOH + Na)

20

Elemental Analysis : Calcd for C ₅₅ H ₇₃ N ₈ NaO ₂₀ • 5H ₂ O			
Found	C : 51.63 C : 51.86	H : 6.53 H : 6.71	N : 8.75 N : 8.86

25

Example 15The Object Compound was obtained according to a similar manner to that of Example 14.IR(Nujol) : 3250, 1600, 1060cm⁻¹

30 NMR(DMSO-d₆, δ) : 0.86(3H, t, J = 6.6Hz), 0.95(3H, d, J = 6.7Hz), 1.09(3H, d, J = 5.9Hz), 1.2-1.53(10H, m), 1.59-2.1(5H, m), 2.13-2.68(4H, m), 2.96(1H, m), 3.0-3.65(2H, m), 3.7-4.55 (16H, m), 4.6-5.68(10H, m), 6.40(1H, d, J = 8.1Hz), 6.59(1H, s), 6.61(1H, d, J = 8.1Hz), 6.95(2H, d, J = 8.8Hz), 7.4(1H, d, J = 7Hz), 7.51(1H, d, J = 8Hz), 7.78(brs, 1H), 7.90(2H, d, J = 8.8Hz), 8.13(1H, d, J = 7Hz), 8.35-8.6(2H, m), 8.68(1H, s), 8.73(1H, s)

35



40

Example 16

45 To a suspension of Starting Compound (1.0g) and triethylsilane (0.94g) in dichloromethane (5.0ml) was gradually added trifluoroacetic acid (10ml) and stirred for 30 minutes under nitrogen atmosphere. The reaction mixture was evaporated under reduced pressure. To the residue was added pH 6.86 phosphate-buffer, and adjusted to pH 8.5 with 1N sodium hydroxide and subjected to column chromatography on ODS (YMC-gel ODS-AM S-50) and eluted with 30% acetonitrile aq. The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove acetonitrile. The residue was lyophilized to give Object Compound (0.61g).

IR(KBr) : 3300, 1664, 1631, 1444, 1270, 1247, 1047 cm⁻¹

50 NMR (DMSO-d₆, δ) : 0.89(3H, t, J = 7.0Hz), 0.96(3H, d, J = 6.8Hz), 1.09(3H, d, J = 6.3Hz), 1.3-1.5(6H, m), 1.6-2.4(10H, m), 2.9-3.3(2H, m), 3.4-3.5(1H, m), 3.7-4.1(7H, m), 4.1-5.6(8H, m), 4.7-5.3(7H, m), 5.40(2H, d, J = 7.0Hz), 6.7-7.2(7H, m), 7.23(1H, s), 7.42(1H, d, J = 8.5Hz), 7.5-7.8(6H, m), 7.94(2H, d, J = 8.4Hz), 8.10(1H, d, J = 7.0Hz), 8.64(1H, d, J = 7.0Hz), 8.72(1H, s)

FAB-MS : e/z = 1229(M⁺ + Na)

Elemental Analysis : Calcd for C ₅₄ H ₇₁ N ₈ NaO ₂₀ S•6H ₂ O			
Found	C : 49.31 C : 49.38	H : 6.40 H : 6.36	N : 8.61 N : 8.52

5

Example 17

The Object Compound was obtained according to a similar manner to that of Example 11.
 10 NMR(DMSO-d₆ δ) : 0.86(3H, t, J = 6.8Hz), 0.96(3H, d, J = 6.7Hz), 1.09(3H, d, J = 6.0Hz), 1.2-1.5(12H, m), 1.6-2.5(8H, m), 2.9-3.5(4H, m), 3.7-4.1(5H, m), 4.1-4.5(8H, m), 4.7-4.5(7H, m), 6.7-7.1(6H, m), 7.24(1H, s), 7.40(1H, d, J = 8.4Hz), 7.5-7.8(6H, m), 7.94(2H, d, J = 8.4Hz), 8.12(1H, d, J = 8.4Hz), 8.61(1H, d, J = 7.7Hz), 8.72(1H, s),
 15 FAB-MS : e/z = 1271.1(M⁺ + Na)

15

Elemental Analysis : Calcd for C ₅₇ H ₇₇ N ₈ NaO ₂₀ S•7H ₂ O			
Found	C : 49.78 C : 49.94	H : 6.67 H : 6.80	N : 8.15 N : 8.16

20

Example 18

The Object Compound was obtained according to a similar manner to that of Example 11.
 25 IR(KBr-Pelet) : 1631, 1537, 1515, 1494, 1440, 1245, 1045, 804cm⁻¹
 NMR(DMSO-d₆, δ) : 0.88-1.07(15H, m), 1.10-1.18(4H, d, J = 6.3Hz), 1.2-1.4(2H, m), 1.5-1.9(2H, m), 2.2-2.4(2H, m), 2.9-3.1(1H, m), 3.1-3.3(2H, m), 3.7-3.9(1H, m), 3.9-4.1(6H, m), 4.2-4.6(7H, m), 5.7-5.8(1H, m), 4.90(2H, d, J = 6.4Hz), 5.0-5.1(1H, m), 5.1-5.2(4H, m), 5.3-5.4(1H, m), 6.7-6.8(2H, m), 6.86(1H, s), 6.99(2H, 30 s), 7.03(2H, d, J = 8.8Hz), 7.22(1H, s), 7.4-7.5(1H, m), 7.6-7.8(6H, m), 7.95(3H, d, J = 8.8Hz), 8.2(1H, m), 8.6-8.7(1H, d, J = 8.2Hz), 8.76(1H, s)
 FAB-MS : e/z = 1229(M⁺ + Na)

35

Elemental Analysis : Calcd for C ₅₄ H ₇₁ N ₅ NaO ₂₀ S•6H ₂ O			
Found	C : 49.31 C : 49.12	H : 6.36 H : 6.22	O : 8.52 O : 8.48

40

Example 19

To a suspension of Starting Compound(10g) and sodium cyanoborohydride (1.25 g) in dichloromethane-(100ml) was gradually added tetrahydrofuran(50ml) at 5°C. The mixture was stirred at 5°C for 1hour. The reaction mixture was evaporated under reduced pressure. To the residue was added water, adjusted to pH 45 8.5 with 1N sodium hydroxide and subjected to column chromatography on ODS (YMCgel ODS-AM S-50) and eluted with 60%acetonitrile aq. The fractions containing the crude product were combined and evaporated under reduced pressure to remove acetonitrile. The residue was lyophilized to give crude powder(9.7g). The crude powder (1.5g) was purified by preparative HPLC utilizing a C₁₈ Bondapak resin powder(9.7g). The crude powder (1.5g) was purified by preparative HPLC utilizing a C₁₈ Bondapak resin (Waters Associates, Inc.) which was eluted with a solvent system comprised of acetonitrile pH7 phosphate buffer (40:60) at a flow rate of 100ml/min using a Shimazu LC-8A pump. The column was monitored by a UV detector set at 240nm. The fractions containing the major compound at retention time of 24.1 minute were combined and evaporated under reduced pressure to remove acetonitrile. The residue was adjusted to pH8.5 with 1N sodium hydroxide and subjected to column chromatography on ODS(YMC-gel ODS-AM S-50) and eluted with 60% acetonitrile aq. The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove acetonitrile. The residue was lyophilized to give major Object Compound(0.77g). The minor Object Compound (98mg) at retention time of 29.86min. was obtained according to a similar manner to that of the major Object Compoud.

Object Compound (major compound)

NMR (DMSO-d₆, δ) : 0.88(3H, t, J 6.7Hz), 0.96(3H, d, J= 6.7Hz), 1.11(3H, d, J = 5. 8Hz), 1.2-1.6(8H, m), 1.62-2.13(5H, m), 2.13-2.62(4H, m), 2.97(1H, m), 3.18(1H, m), 3.40(1H, m), 3.74(2H, m), 3.83-4.63(14H, m), 4.63-5.40(9H, m), 6.73(1H, d, J = 8. 2Hz), 6.81(1H, dd, J = 1.7, and 8.2Hz), 6.89(1H, s), 7.03(2H, d, J = 8.9Hz), 7.06(1H, d, J = 1.7Hz), 7.22(1H, s), 7.41(2H, m), 7.6-7.8(1H, m), 7.67(2H, d, J = 8.9Hz), 7.71 (2H, d, J = 8.4Hz), 7.93(2H, d, J = 8.4Hz), 8.05(1H, d, J = 8Hz), 8.61(1H, d, J = 6.7Hz), 8.84(1H, s)

IR (Nujol) : 3300, 1625, 1240, 1045cm⁻¹

FAB-MS : e/z = 1259(M⁺ + Na)

10

Elemental Analysis : Calcd : for C ₅₅ H ₇₃ N ₈ NaO ₂₁ S • 6H ₂ O			
Found :	C ; 49.10 C ; 49.02	H ; 6.37 H ; 6.32	N ; 8.33 N ; 8.34

15

Object Compound (minor compound)

IR (Nujol) : 3300, 1620, 1245 cm⁻¹

NMR (DMSO-d₆, δ) : 0.88 (3H,t,J = 6.7Hz), 0.97 (3H,d,J = 6.7Hz), 1.09 (3H,d,J = 5.8Hz), 1.18-1.52 (8H,m), 1.6-2.59 (10H,m), 2.97 (1H,m), 3.18 (1H,m), 3.40 (1H,m), 3.65-4.59 (15H,m), 4.65-5.45 (9H,m), 6.71 (1H,d,J = 8.1Hz), 6.76 (1H,dd,J = 1.7 and 8.1Hz), 6.87 (1H,s), 6.97 (1H,d,J = 1.7Hz), 7.03 (2H,d,J = 8.8Hz), 7.21 (1H,s), 7.41 (1H,d,J = 7.6Hz), 7.60 (1H,m), 7.63-7.8 (1H,m), 7.67 (2H,d,J = 8.8Hz), 7.71 (2H,d,J = 8.4Hz), 7.94 (2H,d,J = 8.4Hz), 8.10 (1H,d,J = 8Hz), 8.59 (1H,d,J = 8Hz), 8.72 (1H,s)

25

FAB-MS : e/z = 1243 (M⁺ + Na)

30

Elemental Analysis : Calcd : for C ₅₅ H ₇₃ N ₈ NaO ₂₀ S • 7H ₂ O			
Found :	C ; 49.03 C ; 49.24	H ; 6.51 H ; 6.25	N ; 8.32 N ; 8.44

Example 20

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To a mixture of Starting Compound (1g) and mercaptoacetic acid (290 μ l) was added trifluoroacetic acid (10ml), and the mixture was stirred at room temperature for 16 hours under nitrogen atmosphere. The mixture was evaporated under reduced pressure. The residue was poured into water (10ml), and filtrated. The precipitate was subjected to column chromatography on silica gel using dichloromethane /acetic acid/ methanol/water(3:1:1:1) as eluent. The fractions containing the Object Compound was combined and evaporated under reduced pressure. The residue was adjusted to pH8. 0 with 1N sodium hydroxide and subjected to column chromatography on ODS (YMC-gel ODS-AMS50) and eluted with CH₃CN-H₂O (60-40). The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove acetonitrile. The residue was lyophilized to give Object Compound (515mg).

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IR (Nujol) : 3300, 1662, 1627, 1203 cm⁻¹

FAB-MS : e/z = 1205 (M⁺)

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Elemental Analysis : Calcd for C ₅₅ H ₇₃ N ₈ NaO ₁₉ S • 5H ₂ O			
Found	C ; 51.00 C ; 51.15	H ; 6.46 H ; 6.35	N ; 8.65 N ; 8.56

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Example 21

The Object Compound was obtained according to a similar manner to that of Example 20.

IR (Nujol) : 3300, 1672, 1658, 1629, 1530, 1438, 1270, 1214cm⁻¹

FAB-MS : e/z = 1220 (M⁺ + 1)

Elemental Analysis : Calcd for C ₅₆ H ₈₇ N ₈ NaO ₂₅ S • 6H ₂ O			
Found	C ; 50.67 C ; 50.67	H ; 6.61 H ; 6.47	N ; 8.44 N ; 8.38

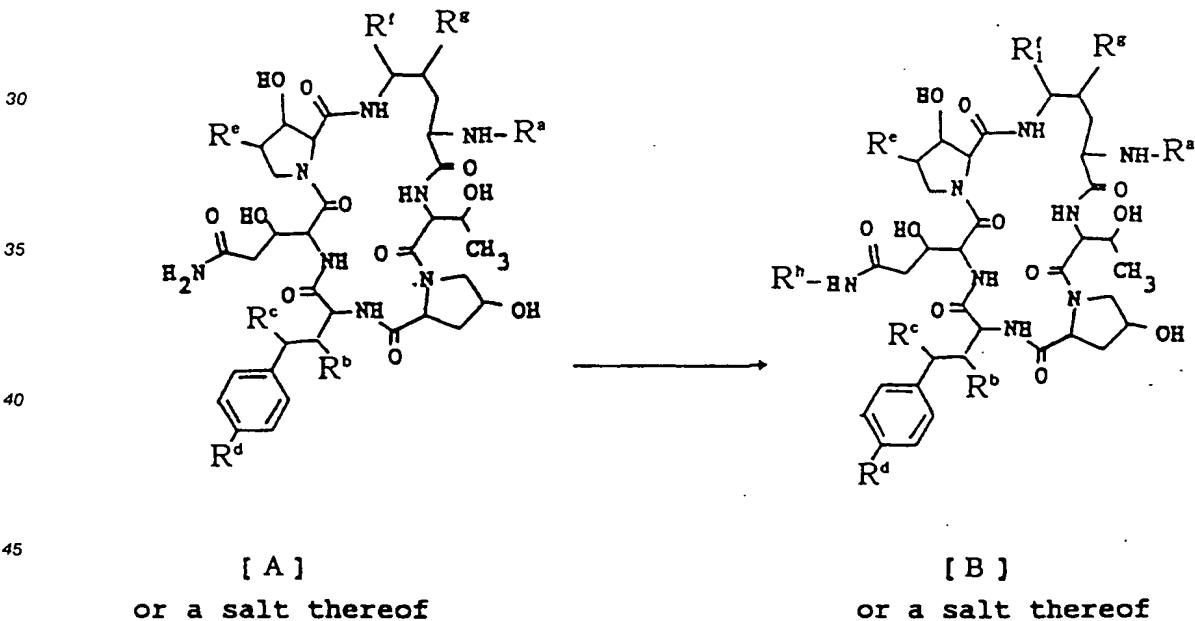
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Example 22

To a solution of Starting Compound (2g) in dimethyl formamide (20ml) in presence of Molecular Sieves 4 Å was added glyoxylic acid (0.744g) at ambient temperature. The mixture was stirred for 7 hours at the same temperature. The reaction mixture was pulverized with ethyl acetate. The precipitate was collected by filtration and dried under reduced pressure. The powder was added to water and adjusted to pH7.5. The solution was subjected to column chromatography on ODS (YMC-gel ODS-AM S-50) and eluted with 20 % acetonitrile aq. The fractions containing the Object Compound were combined and evaporated under reduced pressure to remove acetonitrile. The residue was lyophilized to give Object Compound (1.26g).
 IR (KBr, Pellet) : 3350, 1658, 1631, 1248cm⁻¹
 NMR (DMSO-d₆, δ) : 0.88 (3H, t, J = 6.6Hz), 0.95 (3H, d, J = 6.6Hz) 1.18 (3H, d, J = 5.5Hz), 1.23-1.55 (8H, m.), 1.55-2.6 (9H, m), 3.09 (1H, m), 3.1-3.5 (2H, m), 3.6-5.8 (27H, m), 6.74 (1H, d, J = 8.2Hz), 6.83 (1H, d, J = 8.2Hz), 7.03 (2H, d, J = 8.8Hz), 7.05 (1H, s), 7.23-7.55 (3H, m), 7.68 (2H, d, J = 8.7Hz), 7.72 (2H, d, J = 8.4Hz), 7.96 (2H, d, J = 8.4Hz), 8.0-8.2 (2H, m), 8.45 (1H, m), 8.84 (1H, s)
 FAB-MS : e/z = 1259 (M⁺ + Na)

Incidentally, it is to be noted that the known compound [A] of the following formula or a salt thereof can be converted to the compound [B] of the following formula or a salt thereof according to the similar manners to those of the Processes 1 and 5.

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50 where in

R^a is acyl group,
 R^b, R^c, R^f, and R^g are each independently hydrogen or hydroxy,
 R^d is hydroxy, acyloxy, phosphoryloxy or sulfonyloxy,
 R^e is hydrogen or methyl,
 R^f is hydrogen or lower alkoxy, and
 R^g is lower alkyl which may have one or more suitable substituent (s),

55 The compound [A] was disclosed, for example, in Japanese laid-open No. 1-163179, 3-163096, 4-217694, 4-217695, 4-217696, 4-217697, 4-217698, and 4-217699.

Claims

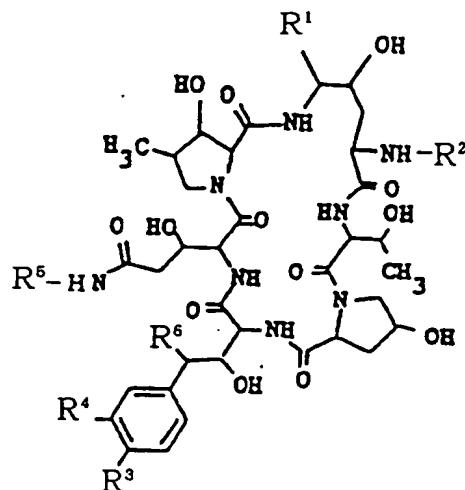
1. A polypeptide compound of the following general formula :

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[I]

wherein

25 R¹ is hydrogen,
 R² is acyl group,
 R³ is hydroxy or acyloxy,
 R⁴ is hydroxy or hydroxysulfonyloxy,
 R⁵ is hydrogen or lower alkyl which may have one or more suitable substituent (s), and
 R⁶ is hydrogen, hydroxy or acyl (lower) alkylthio and a pharmaceutically acceptable salt thereof.

2. A compound of claim 1, wherein

R² is aroyl which may have one or more i) lower alkoxy, ii) higher alkoxy or iii) aryl which may have one or more lower alkoxy or higher alkoxy, higher alkanoyl, and
 35 R⁵ is hydrogen or lower alkyl which may have one or more suitable substituent (s) selected from the group consisting of hydroxy, acyl, di (lower) alkyl amino and cyclic amino.

3. A compound of claim 2, wherein

R² is benzoyl or naphthoyl, each of which may have 1 to 3 suitable substituent (s) selected from the group consisting of lower alkoxy ; higher alkoxy ; and phenyl which may have 1 to 3 higher alkoxy ;
 R³ is hydroxy,
 R⁵ is hydrogen, and
 R⁶ is hydrogen or hydroxy.

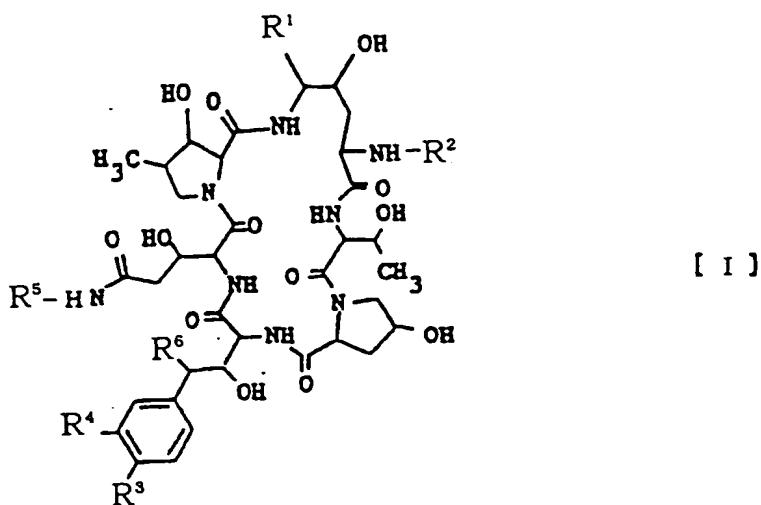
45 4. A compound of claim 3, wherein

R² is benzoyl having higher alkoxy, benzoyl having phenyl which has higher alkoxy, naphthoyl having lower alkoxy, or naphthoyl having higher alkoxy, and
 R⁶ is hydrogen.

50 5. A compound of claim 3, wherein

R² is benzoyl having higher alkoxy, benzoyl having phenyl which has higher alkoxy, naphthoyl having lower alkoxy, or naphthoyl having higher alkoxy, and
 R⁶ is hydroxy.

55 6. A process for the preparation of a polypeptide compound of the formula [I] :



wherein

R¹ is hydrogen,

R² is acyl group,

R³ is hydroxy or acyloxy,

R⁴ is hydroxy or hydroxysulfonyloxy,

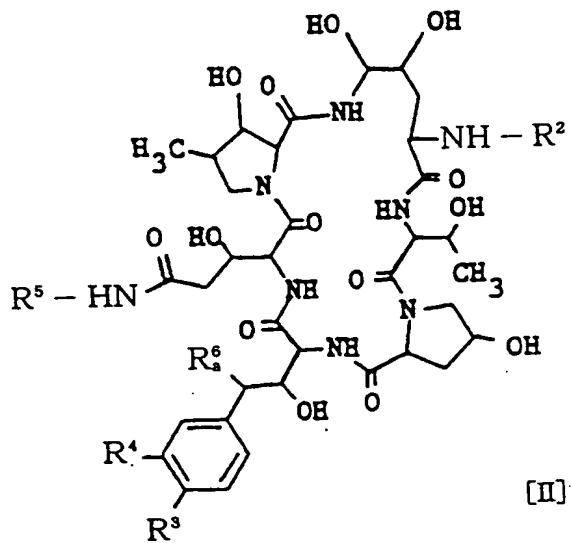
R⁵ is hydrogen or lower alkyl which may have one or more suitable substituent (s), and

R⁶ is hydrogen, hydroxy or acyl (lower) alkylthio,

or a pharmaceutically acceptable salt thereof,

which comprises

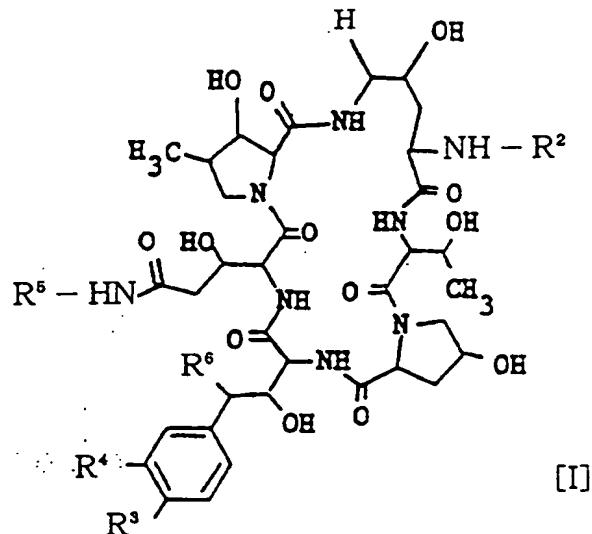
1) reducing a compound of the formula :



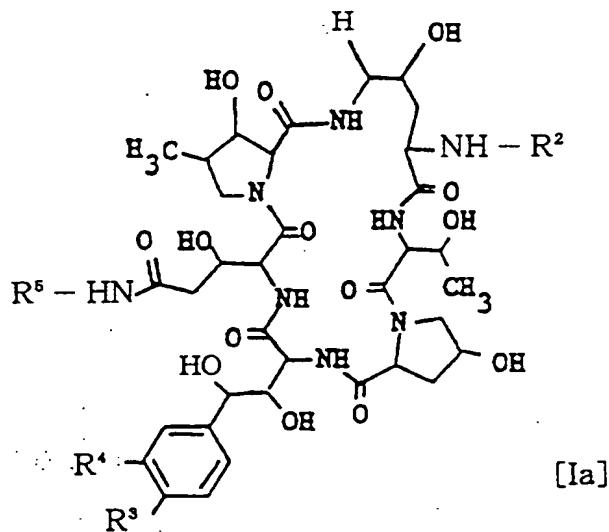
wherein R², R³, R⁴ and R⁵ are each as defined above,

R⁶ is hydroxy or acyl (lower) alkylthio,

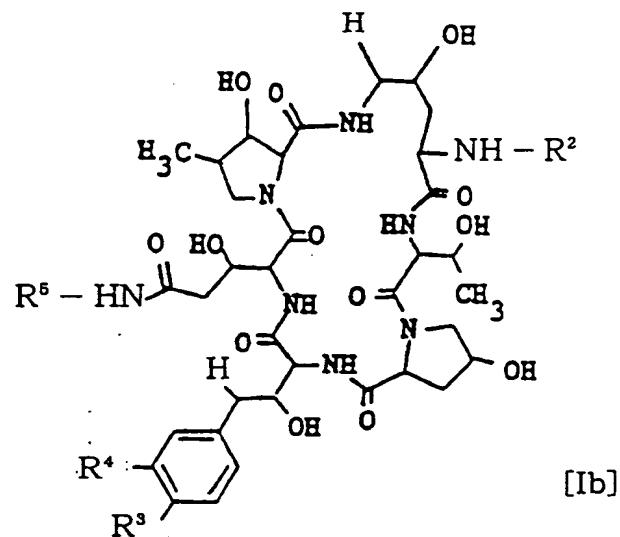
or a salt thereof, to give a compound [I] of the formula :



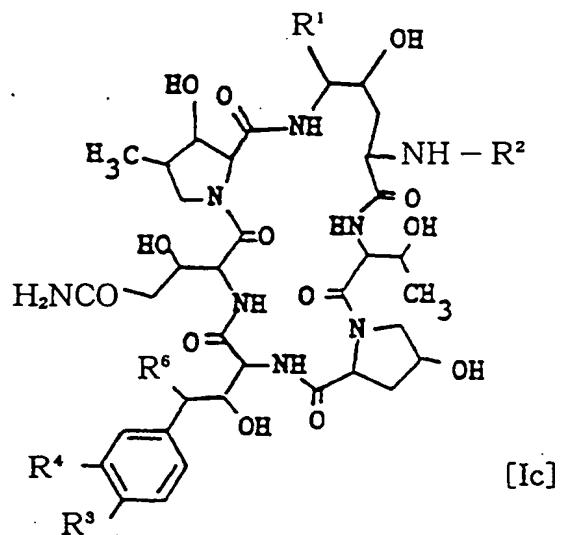
wherein R^2 , R^3 , R^4 , R^5 and R^6 are each as defined above, or a salt thereof, or
(2) reducing a compound of the formula :



wherein R^2 , R^3 , R^4 and R^5 are each as defined above, or a salt thereof; to give a compound [Ib] of the formula :



wherein R², R³, R⁴ and R⁵ are each as defined above, or a salt thereof, or
(3) subjecting a compound of the formula :



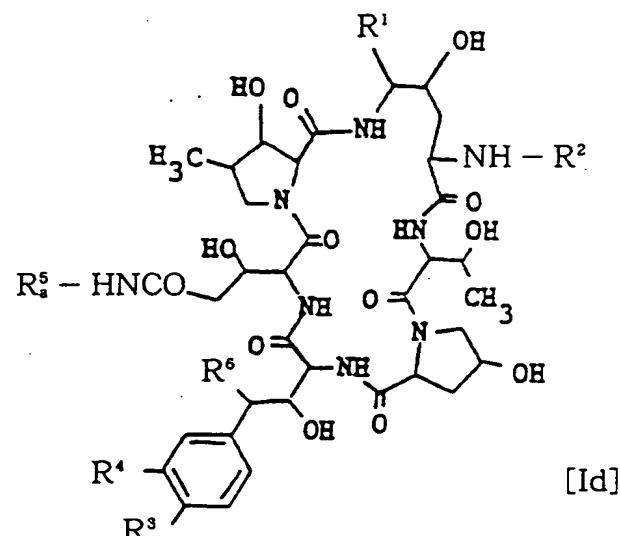
wherein R¹, R², R³, R⁴ and R⁶ are each as defined above, or a salt thereof, to alkylation reaction, to give a compound [Id] of the formula :

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Wherein R¹, R², R³, R⁴ and R⁶ are each as defined above ,

R⁵ is lower alkyl which may have one or more suitable substituent (s), or a salt thereof, or

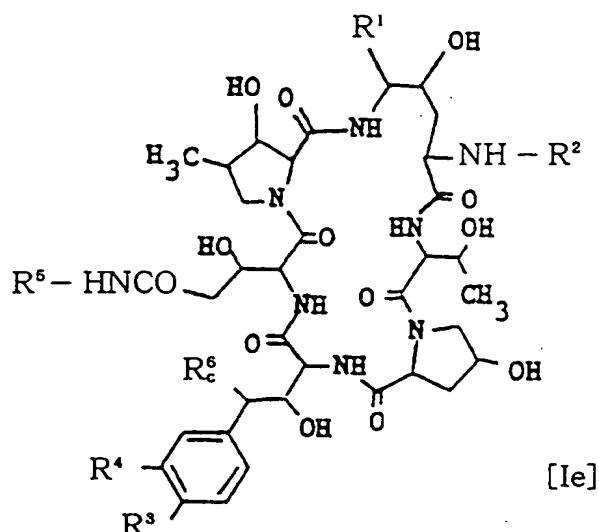
25 (4) subjecting a compound of the formula :

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Wherein R¹, R², R³, R⁴, and R⁵ are each as defined above ,

R⁶ is hydroxy,

or a salt thereof, to substitution reaction, to give a compound [If] of the formula :

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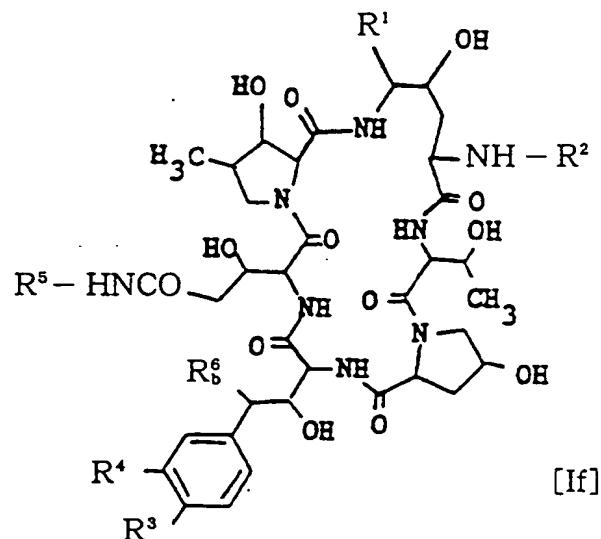
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Wherein R¹, R², R³, R⁴, and R⁵ are each as defined above ,

R⁶ is acyl (lower) alkylthio,

or a salt therof, or

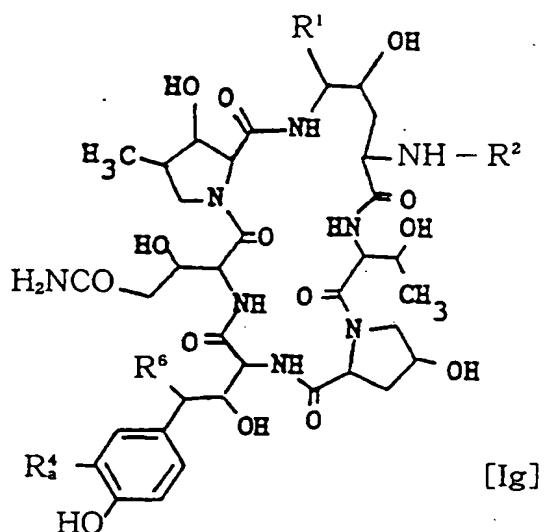
(5) subjecting a compound of the formula :

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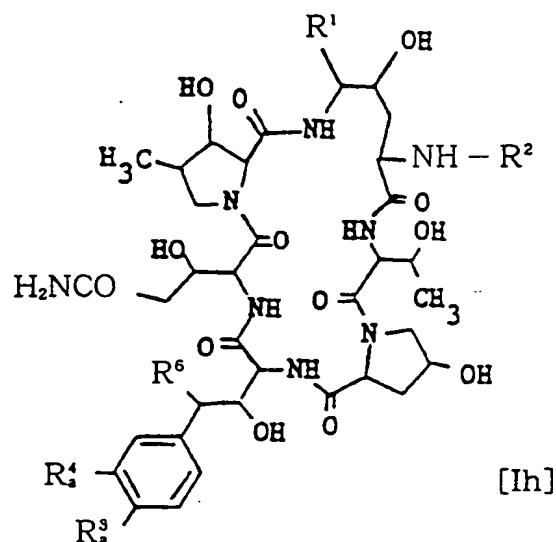
Wherein R¹, R² and R⁶ are each as defined above ,

R⁴ is hydroxysulfonyloxy,

or a salt thereof, to acylation reaction, to give a compound [Ih] of the formula :

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Wherein R¹, R², R⁴_a and R⁶ are each as defined above,
 R³_a is acyloxy,
 or a salt thereof.

7. A pharmaceutical composition which comprises, as an active ingredient, a compound of claim 1 or a pharmaceutically acceptable salt thereof in admixture with a pharmaceutically acceptable carrier or excipient.
8. Use of a compound of claim 1 or a pharmaceutically acceptable salt thereof for the manufacture of a medicament, preferably for the prophylactic and/or the therapeutic treatment of infectious diseases caused by pathogenic microorganism.
9. A compound of claim 1 or a pharmaceutically acceptable salt thereof for use as a medicament.
10. A compound of claim 1 or a pharmaceutically acceptable salt thereof for use in the prophylactic and/or the therapeutic treatment of infectious diseases caused by pathogenic microorganism.



European Patent
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EUROPEAN SEARCH REPORT

Application Number
EP 94 10 7406

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.)
Y	EP-A-0 462 531 (FUJISAWA) 27 December 1991 * examples 1-55; claims * ----	1-10	C07K7/56 C07K1/00 A61K38/12
Y	EP-A-0 448 354 (MERCK & CO.) 25 September 1991 * pages 4-8 , compounds A-2, Z-2, and Z-4; claims * -----	1-10	
TECHNICAL FIELDS SEARCHED (Int.Cl.)			
C07K A61K			
The present search report has been drawn up for all claims			
Place of search	Date of completion of the search		Examiner
MUNICH	24 January 1995		Hermann, R
CATEGORY OF CITED DOCUMENTS			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	